



Mass fluxes and clay mineral formation in soils developed on slope deposits of the Kowarski Grzbiet (Karkonosze Mountains, Czech Republic/Poland)

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Abstract: Weathering, mineral formation, and transformation processes along slopes are complex. In cool mountainous regions, undisturbed soil development with a strong vertical leaching element may abruptly change as a result of erosion, accumulation, lateral water fluxes and aeolian input. We investigated soils in the eastern Karkonosze Mountains that have developed on silicatic slope deposits. To date, illite, vermiculite and chlorite are the minerals that have been detected in the clay fraction. Although the climate and parent material should be favourable for the formation of smectites, expandable phases were not verified so far. We investigated if expandable phases could be detected and how they related to elemental fluxes along a short slope sequence (1142–1268 m a.s.l. on the border between the Czech Republic and Poland). Mass balance calculations indicated intensive mineral weathering together with a significant leaching of Mg, Al, Ca and Mn on the shoulder and foot slope positions. In the middle zone, which has a concave or undulating surface shape, however, the mass balances of several elements (Na, K, Al, P) revealed a less pronounced leaching (corresponding to a lower degree of podsolization) and in some cases even accumulation. At all sites, mass balance calculations and detected soil minerals (e.g. the increase in illite towards the surface together with an increase in Al and K) indicate some aeolian input. Kaolinite was detected in all soil horizons. Its concentration slightly increased towards the soil surface. Together with the pronounced leaching of Ca, part of kaolinite originates from plagioclase weathering. Besides being a weathering product of primary minerals, part of the kaolinite is inherited from the parent material and probably is also due to aeolian input. In all soils, illite was being transformed into vermiculite and smectites (through regularly-interstratified illite-smectite phases). In addition, the content of chloritic components which increases with depth indicated their concurrent weathering and transformation into smectites. Amphibole also may have acted as a source of smectites. Not all smectite is being actively formed in the soil. Most likely due to slope processes (cover beds) that affected even the subsoil, some smectite has been transferred along the slope. Part of the smectite also seems to derive from the parent material. Active formation of expandable clay minerals is related to convex and planar landscape forms. This relationship suggests intense element leaching, inheritance from the parent material and cover bed mixing processes have contributed to the presence of smectite. Along the slope, zones with predominant vertical transport (shoulder, foot slopes) may repeatedly be interchanged with zones dominated by lateral transport (undulating slope, concave forms).

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Abstract

Weathering, mineral formation, and transformation processes along slopes are complex. In cool mountainous regions, undisturbed soil development with a strong vertical leaching element may abruptly change as a result of erosion, accumulation, lateral water fluxes and aeolian input. We investigated soils in the eastern Karkonosze Mountains that have developed on silicatic slope deposits. To date, illite, vermiculite and chlorite are the minerals that have been detected in the clay fraction. Although the climate and parent material should be favourable for the formation of smectites, expandable phases have not so far been verified. We investigated if expandable phases could be detected and how they related to elemental fluxes along a short slope sequence (1142 – 1268 m a.s.l. on the border between the Czech Republic and Poland). Mass balance calculations indicated intensive mineral weathering together with a significant leaching of Mg, Al, Ca and Mn on the shoulder and foot slope positions. In the middle zone, which has a concave or undulating surface shape, however, the mass balances of several elements (Na, K, Al, P) revealed a less pronounced leaching (corresponding to a lower degree of podsolization) and in some cases even

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26 accumulation. At all sites, mass balance calculations and detected soil minerals (e.g. the increase in
27 illite towards the surface together with an increase in Al and K) indicate some aeolian input.
28 Kaolinite was detected in all soil horizons. Its concentration slightly increased towards the soil
29 surface. Together with the pronounced leaching of Ca, part of kaolinite originates from plagioclase
30 weathering. Besides being a weathering product of primary minerals, part of the kaolinite is
31 inherited from the parent material and is probably also due to aeolian input. In all soils, illite has
32 being transformed into vermiculite and smectites (through regularly-interstratified illite-smectite
33 phases). In addition, the content of chloritic components which increases with depth indicated their
34 concurrent weathering and transformation into smectites. Amphibole may also have acted as a
35 source of smectites. Not all smectite is actively formed in the soil. Most likely due to slope
36 processes (cover beds) that even affected the subsoil, some smectite has been transferred along the
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38 expandable clay minerals is related to convex and planar landscape forms. This relationship
39 suggests that intense element leaching, inheritance from the parent material and cover bed mixing
40 processes have contributed to the presence of smectite. Along the slope, zones with predominant
41 vertical transport (shoulder, foot slopes) may be repeatedly interchanged with zones dominated by
42 lateral transport (undulating slopes, concave forms).

43

44 **1. Introduction**

45 Clay minerals are good indicators of former and current weathering processes in mountain
46 soils (Reynolds, 1972; Righi et al., 1999; April et al., 2004). Egli et al. (2003a) showed that an
47 understanding of mineral formation, transformation, and weathering rates is fundamental in the
48 analysis of the reactivity of fragile mountain ecosystems. Clays are often used for estimating the
49 degree of soil development (Certini et al., 1998; Fichter et al., 1998; Mirabella and Sartori, 1998;
50 Mirabella et al., 2002; Bonifacio et al., 2009; Mavris et al., 2015). Egli et al. (2001) and Mavris et

51 al. (2010) verified the applicability of clay mineralogy to track the initial stages of soil formation in
52 proglacial areas. Kabala and Zapart (2012) demonstrated the usefulness of clay minerals to
53 investigate the weathering intensity and the development of a *cambic* horizon formation on
54 moraines of an Arctic glacier. Clay minerals and soil minerals can also be used as pedosignatures
55 for linking soil evolution and landscape development (Mourier et al., 2010; Böhlert et al., 2011;
56 Dahms et al., 2012; Jäger et al., 2015).

57 Studies of Podzols that developed from acid crystalline rocks in mountain environments
58 have evidenced the presence of vermiculite, illite/vermiculate, illite/smectite, hydroxyl-Al-
59 interlayered vermiculite (HIV) and chlorites (Wilson, 1986; Bain et al., 1990; Dahlgren and
60 Ugolini, 1991; Egli et al., 2003; Kitagawa, 2005; Viennet et al., 2015). Several authors have
61 described different mechanisms of clay mineral formation associated with the podzolization
62 process, including chlorite to smectite transformation (Wilson, 1999) or smectite and mixed-layered
63 mineral formation at the expense of biotite (Mavris et al., 2011). In addition, imogolite and poorly
64 crystalline phases such as allophone formation have frequently been described (Wada, 1989;
65 Ugolini and Dahlgren, 1991). Most of these studies were conducted in soils developed on more or
66 less homogeneous parent materials (Righi et al., 1999, Egli et al., 2008); whereas in Central
67 European low-/mid-mountains soils are often either polygenetic (Waroszewski et al., 2013, Geitner
68 et al., 2014) or developed from stratified slope deposits (Migoń and Kacprzak, 2014; Waroszewski
69 et al., 2015a, Waroszewski et al., 2015b). Selective and non-selective slope transport produces a
70 heterogeneous sediment differing in petrography and texture which directly or indirectly affects
71 soil formation, morphology and soil-plant interactions (Kleber and Terhorst, 2013); thus, this may
72 also influence the weathering rates, clay formation and transformation processes.

73 Illite has traditionally been recognized as a main component of the clay fraction in soils of
74 the Karkonosze Mountains (Szerszeń, 1974; Bogda, 1981; Bogda et al., 1998). Also, biotite to
75 vermiculite transformation has been described (Kowalinski et al., 1967; Kabala, 2005). The

76 presence of smectite has, however, not been confirmed, although the weathering environment, such
77 as strong elemental leaching (Kitagawa, 2005), would be favourable for its formation and precursor
78 minerals such as mica are present. The clay mineralogy of the mountain soils of Central Europe,
79 particularly on slope covers, is still poorly investigated (Kacprzak and Derkowski, 2007).

80 With this investigation, we attempt to improve the understanding of soil and clay mineral
81 formation in these environments and to answer the question as to under which conditions smectites
82 might or might not be encountered. As such, we aimed at investigating mass fluxes and the
83 formation/transformation of clay minerals in soils under different leaching conditions within a slope
84 on acidic metamorphic rocks (mica schists). A particular focus was given to the detection of
85 smectites and their relation to soil evolution along a toposequence.

86

87 **2. Study area and sampling sites**

88 The study was carried out on the Czech-Polish border, in the Karkonosze Mountains, the
89 highest part of the Sudeten Mountains (Fig.1). The main ridge (1400 – 1450 m a.s.l.) of the
90 Karkonosze Mts is a plateau with a WNW-ESE orientation. The northern part of the Karkonosze
91 Mts borders with the Jelenia Góra Basin (350 – 500 m a.s.l.) and the southern part is bounded by
92 deep river valleys (Engel et al., 2010; Migoń, 1999). The Karkonosze Mountains consist of
93 Carboniferous granite in their central part, whereas the outer parts are built of metamorphic Pre-
94 Cambrian/Old-Paleozoic metamorphic rocks (gneiss, mica schists, amphibolite schists, etc.). The
95 current relief of the Karkonosze Mountains is the result of weathering and denudation during the
96 Tertiary and the Pleistocene glaciations and related mass movements. The Pleistocene Scandinavian
97 ice-sheets reached the foothills of the Sudeten Mts on two occasions: during the San II (Mindel,
98 Pre-Illinoian; MIS stage 12) and Odra (Riss; Illinoian; MIS stage 6) glaciations. During that time,
99 the permanent snow cover was at lower altitudes which promoted snow accumulation in the glacial
100 cirques; as a consequence, mountain glaciers started to develop. The last glacial maximum (LGM)

101 and late glacial oscillations (Younger Dryas) of the mountain glaciers in the Karkonosze Mountains
102 correlate with the glaciations of the Alps and the Bavarian/Bohemian Forest (Engel et al., 2014).
103 Severe and long-lasting periglacial conditions during the Pleistocene favoured physical weathering
104 and cryogenic slope processes (Migoń and Traczyk, 2000).

105 The spatial distribution of soils in the Karkonosze Mountains is a function of altitude
106 (climate) and geology resulting in a clear vertical soil zonality. Haplic/Albic Luvisols are present up
107 to 750 m a.s.l., usually on the fine-textured colluvia. The zone between 500 and 800 m a.s.l. is
108 characterized by Dystric/Alumic Cambisols (usually skeletal) developed from granite or gneiss
109 regolith. In the upper forest zone (800 – 1000 m a.s.l.), Albic/Entic Podzols prevail with a high
110 content of rock fragments. Depending on the surface relief between 1000 and 1200 m a.s.l., Podzols
111 become either (1) extremely skeletal when developed on former rock glaciers and block covers
112 (Hyperskeletal Podzols) or (2) increasingly wet if situated on plateaus (Stagnic/Histic Podzols).
113 Histic Podzols and Histosols dominate the zone between 1200 and 1400 m a.s.l. Above an altitude
114 of 1400 m a.s.l., only Lithic Leptosols, Hyperskeletal Leptosols and Skeletal Ragosols can be found
115 (Kabala et al., 2013).

116 The sampling sites were located in the Kowarski Grzbiet (Střeha), which belongs to the
117 Kowary metamorphic unit (eastern Karkonosze Mts, Aleksandrowski et al., 2013). This area has
118 never been covered by continental or mountain glaciers. However, a strong influence of periglacial
119 conditions was assumed (Jahn, 1963). The Kowarski Grzbiet (Střeha) is a 4-km-long ridge,
120 asymmetrically lifted with an almost completely aligned plateau. It is made of gneiss and mica
121 schist, with local inclusions of chlorite schist, amphibolite schist and quartzite (Aleksandrowski et
122 al., 2013). Waroszewski et al. (2010) recognized that the slope deposits are the parent material for
123 Podzols in the Kowarski Grzbiet. They described remarkable differences in soil morphology as a
124 function of the (1) sandy, hyperskeletal top layer and (2) loamy-skeletal dense basal layer.

125 Mean annual precipitation for this region varies between 900 mm in the foothills, and up to 1350
126 mm on the upper slopes. Mean annual air temperature ranges between 7°C at 600 m a.s.l. and 1 –
127 2°C at 1600 m a.s.l. The duration of snow cover reaches about 200 days per year on the highest
128 peaks (Sobik et al., 2013). The present study was performed in a narrow section of the mountain
129 range (1142 – 1268 m a.s.l.); thus, the climatic conditions along the investigated slope do not
130 include high variability. The vegetation at the sampling sites is dominated by *Picea abies* in the tree
131 layer and *Vaccinium myrtillus* on the forest floor, accompanied by grass species including *Poa*
132 *nemoralis*, *Agrostis capillaries*, *Agrostis canina*, *Poa trivialis*, and *Festuca pratensis*.

133

134 **3. Materials and methods**

135 3.1. Field procedures

136 Four soil profiles were dug along a toposequence (about 450 m long) on a SE-facing slope
137 of the Skalny Stół (1270 m a.s.l., the highest peak of the Kowarski Grzbiet) at an altitude of 1142 –
138 1268 m a.s.l. The site characteristics are presented in Table 1. The soils (Tab. 2) were described
139 according to FAO Guidelines (2006) and the soil reference groups were established using the WRB
140 classification (IUSS working group WRB, 2014). The slope gradient was measured in the field
141 using a clinometer and then rechecked based on the DEM data. Approximately 2 kg of soil material
142 was taken from each soil horizon. Prior to measurements, the samples were dried, gently crushed
143 and passed through a 2 mm sieve. In addition, undisturbed soil samples were collected using steel
144 cylinders to estimate soil bulk density. In some cases (caused by a high skeleton content that
145 disabled the use of cylinders), soil bulk density was determined by excavating holes (and weighing
146 the excavated material) that were backfilled with a measurable volume of quartz sand. To check the
147 geological composition, at least 40 rock fragments per soil profile were crushed to enable a
148 petrographical classification. Stones with a diameter of at least 15 cm were chosen for thin section
149 preparation.

150

151 3.2. Description of rock thin sections

152 Petrographic description of the polished thin sections of dominant rocks was carried out
153 using a Leica DM2500P optical microscope in transmitted and reflected light. The modal
154 composition of the studied rocks was estimated by point counting using the freeware software
155 JMicroVision v1.2.7.

156

157 3.2. Soil analyses

158 After sample pretreatment with 3% H₂O₂ and dispersion with hexametaphosphate, particle-
159 size distribution was performed using sieves (sand fraction) and the hydrometer method for the silt
160 and clay fraction (Van Reeuwijk, 2002). Soil pH was potentiometrically measured in a 1:2.5
161 (soil:distilled water) suspension. Soil organic carbon (SOC) content was determined with the dry
162 combustion method using CO₂ spectroscopic measurements (Ströhlein CS-mat 5500 analyzer).
163 Exchangeable ions (Ca²⁺, Mg²⁺, K⁺, Na⁺) were extracted with 1 M ammonium acetate at pH = 7.
164 Amorphous, poorly crystalline and crystalline Fe and Al forms were determined with the dithionite
165 and oxalate extractions (McKeague et al., 1971). Cation concentrations in the extracts were
166 measured via atomic absorption spectroscopy (AAS) and inductively coupled plasma (ICP-OES).

167 Elemental composition (Si, Al, Ti, Fe, K, Na, Ca, Mg, Mn, P, Cl, and S) was measured
168 using X-ray fluorescence (XRF). Soil material was milled to 50 µm in a tungsten carbide disc swing
169 mill (Retsch® RS1, Germany). Thereafter, the material was weighed into plastic cups with Prolen®
170 foil at the bottom and analysed with an energy dispersive X-ray fluorescence spectrometer
171 (SPECTRO X-LAB 2000, SPECTRO Analytical Instruments, Germany).

172

173 3.3. Mass fluxes

174 The long-term weathering rates of soils were derived from the calculations of
 175 enrichment/depletion factors determined using immobile element contents (Brimhall and Dietrich,
 176 1987; Egli and Fitze, 2000). In this investigation, Ti was used as an immobile element. Although
 177 some studies indicate that Ti might be mobile to some extent in weathering environments (Cornu et
 178 al., 1999), other ‘stable’ elements such as Zr are also criticised due to grain corroding caused by
 179 surface dissolution (Balan et al., 2001). Therefore, we followed the statement of Stiles et al. (2003),
 180 who concluded that Ti is better suited for mass-balance calculations during pedogenesis, as Ti has
 181 higher concentrations and is enriched in the clay fraction. The ratio can be used as a weathering
 182 index: the lower the ratio (Ti/Ca), the higher the degree of weathering (Dahms et al., 2012).

183 Relative elemental losses in the soil column can be calculated using the open-system mass
 184 transport function $\tau_{j,w}$ (Chadwick et al., 1990):

185

$$186 \quad \tau_{j,w} = \left(\frac{C_{j,w} \cdot C_{i,p}}{C_{i,w} \cdot C_{j,p}} \right) - 1 \quad (1)$$

187

188 where: i denotes the immobile element (Ti), $C_{j,p}$ (g/kg) is the concentration of element j in the
 189 unweathered parent material, and $C_{j,w}$ is the concentration of element j in the weathered product
 190 (g/kg).

191

192 3.4. Clay mineralogy

193 The mineralogy of the clay fraction was determined after removal of organic matter and, in
 194 some cases (mostly from Bh and Bhs horizons), iron oxides. Oriented specimens on glass slides
 195 were analysed via X-ray diffraction (XRD) using Cu-K α radiation from 2 to 15°2 θ with steps of
 196 0.02°2 θ at 2 s/step. The following treatments were performed: Mg-saturation, ethylene glycol
 197 solvation (EG), and K-saturation. After an initial XRD scan, the K-saturated samples were heated

for 2 h at 335 and 550°C, and rescanned after each heating step. The $d(060)$ region was studied on randomly oriented samples that were step-scanned from 58 to 64°2 θ with steps of 0.02°2 θ at 10 second intervals using a Bruker AXS D8 Advance (CuK α). In addition, a scan from 2 to 80°2 θ (with steps of 0.02°2 θ) was performed. Digitised X-ray data in the range of 2 to 15°2 θ were corrected for Lorentz and polarisation factors (Moore and Reynolds, 1997). Diffraction patterns in the range of 2 to 15°2 θ and 58 to 64°2 θ were fitted by the Origin™ PFM program using the Pearson VII algorithm. Background values were calculated by means of a non-linear function (polynomial 2nd order function; Lanson, 1997).

To check the presence of kaolinite and imogolite-type materials, DRIFT (Diffuse Reflection Infrared Fourier Transform; Bruker, Tensor 27) measurements were carried out from 250 to 4000 cm⁻¹. About 30 mg of clay material and 270 mg of KBr were homogenised in a mill using a fine ball-mill (Zr) for 30 s and at 10 rpm. Prior to measurement, the samples were dried in an oven at 70°C for 2 hours. The individual spectra were normalised using OPUS 6 software. The FT-IR spectra of soil clays in the OH-stretching region (4000 – 3200 cm⁻¹) were, furthermore, normalised (using the minimum-maximum method) and then compared to each other. This enables the detection of relative changes in the abundance of kaolinite.

4. Results

4.1. Bedrock identification and rock mineralogy

The mica schist had a light grey color and was typically lustrous at the foliation surface. It is present in the coarse skeleton of the soils. The mica schist was characterized by oriented mica crystals and by a lepidogranoblastic texture (Fig. 3). The rock material contained quartz, muscovite and chlorite as its major minerals, whereas the minor and accessory constituents were biotite, garnet, sphene, tourmaline, zircon and opaques. The schists had alternating layers consisting of quartz grains up to a thickness of 0.5 mm and mica flakes up to 1 mm (Fig. 3b). Locally, within the

223 mica layers, a larger amount of chlorite was observed (Fig. 3c and 3d). The schist has experienced
224 some deformation, as evidenced by the clearly visible cleavage in some parts of the rock. Quartz
225 occurred as a major component. It formed anhedral isometric crystals with a size ranging from 0.1
226 to 0.5 mm. Muscovite was present as subhedral flakes occurring almost exclusively within the mica
227 layers. The size of individual muscovite crystals reached 1 mm. Chlorite and biotite occur within
228 the mica layers as subhedral flakes with a size of up to 0.3 mm and 0.4 mm, respectively. Garnets
229 are isometric, showing subhedral porphyroblasts within quartz or mica layers and reach a size of up
230 to 1 mm. Garnet crystals were strongly weathered, fractured and overgrown with chlorite and
231 secondary brown opaque minerals (Fig. 3c, 3d). Sphene was present exclusively within mica layers,
232 where it formed subhedrally elongated crystals up to 0.25 mm in length. Tourmaline euhedral
233 crystals had a size of about 0.3 mm and showed a trigonal cross-section outline and zoning (Fig. 3c,
234 3d). Tourmaline crystals usually had a size of up to 0.3 mm. Opaque minerals occurred as two
235 varieties. Black opaques formed anhedral crystals of approximately 0.1 mm in thickness. Brown
236 opaques represented secondary phases that were formed due to garnet and/or black opaque
237 alteration. Zircon euhedral crystals were smaller than 0.1 mm and visible as dark brown to black
238 pleochroic haloes within biotite flakes.

239 Other rock fragments that were present in the soil profiles had a quartzite composition.
240 Macroscopically, the quartzite varied in color from grey to white. The quartzite was mostly
241 composed of rounded, anhedral, isometric quartz grains of a size between 0.1 to 0.5 mm. The small
242 quartz grains formed a siliceous matrix (Fig. 3a) for the larger (3 to 8 mm) irregular and anhedral
243 crystals of quartz. These larger grains are characterized by an undulatory extinction, a common
244 feature observed in deformed rocks. Small (< 0.01 mm) gas and/or liquid inclusions were common
245 within quartz grains. These inclusions were arranged in more or less parallel lines. Opaques were
246 the only accessory minerals within the quartzite. They formed subhedral to euhedral crystals with a
247 size of up 0.5 mm.

248

249 4.2. Soil characteristics

250 Generally, the thicknesses of organic layers were in the range of 8 – 20 cm (Table 2). The
251 uppermost mineral horizons were bleached due to podzolization, thus having a grey or greyish-
252 brown colour. Partially reducing conditions, related to water stagnation, were found in the E
253 horizons. In profile KH2, subsurface water flow (in May) was evident at the contact of the E to the
254 Bh horizon. The eluvial horizons contained 10 – 30% rock fragments. The thickness of the eluvial
255 horizons varied between 7 and 18 cm without any particular trend along the slope catena. The
256 underlying dark brown Bh/Bhs horizon usually had a subangular, weak to moderate soil structure
257 and a friable consistency. The skeleton content in Bh horizons increased to 50-70%. The lowermost
258 BC horizons were in general more reddish or yellowish and more compacted and friable to firm,
259 compared to adjacent horizons. Soil structure was blocky angular or platy. Rock fragments
260 accounted for up to 90% of the soil mass, and the orientation of the clasts was mostly horizontal.
261 The bulk density of the topsoil horizons varied between 0.67 and 1.25 g cm⁻³. With depth, bulk
262 density increased in all profiles up to 1.39 g cm⁻³ (BC horizon).

263 In general, all soils were acidic throughout the profile (Table 3). Organic carbon was found
264 to relatively great depths and indicated partially intense translocation (podzolisation) processes. The
265 topsoil layers, except the AE horizon in profile KH3, were strongly depleted in oxalate-extractable
266 Fe and Al. Amorphous iron (Fe_o) was a main fraction of the whole pedogenic forms, as exemplified
267 by the Fe_o/Fe_d ratio, which oscillated within the range of 0.52 and 0.97 (Table 3). Although the soils
268 are located on the same slope and developed in a similar altitudinal (climate) zone and parent
269 material, some differences in particle size distribution were evident. The soil material seems to be
270 relatively homogenous throughout the profiles of KH1 and KH2, with 47 – 52% in the sand and 44
271 – 50% in the silt fraction. A more variable texture, evidenced by a change in the fine silt and fine
272 sand fractions, was found in profile KH3 (Table 4). Profile KH4 had a relatively uniform particle

size distribution with a sandy loam texture throughout the profile. However, it clearly contains less silt (26-38%) and more fine and very fine sand compared to the profiles in the upper-slope (Table 4).

The investigated Podzols and Cambisols developed on slope materials (cover beds) that showed some variations in their coarse fragments and silt content. The upper layers were characterized by a lower rock clast content (10 – 30%) and a silt content of up to 50%, while the basal layer had a significantly higher share of skeleton (60 – 90%) and lower silt content (Table 4). A special case is the topsoil of profile KH3, where silt reached 68%; such an enrichment could be related to erosion (near-surface) and accumulation processes.

4.3. Geochemistry and mass fluxes

The elemental composition of the soils reflects the presence of mica schists in the bedrock. Consequently, the SiO_2 (45.9 – 77.0%) and Al_2O_3 (11.6 – 18.3%) content is high. The eluvial horizons are enriched in Al, Na and K, and depleted in Mg, Ca, Mn and Fe. In particular, the Fe content is around 1.5% in the E horizons and up to 11% in the Bh/Bhs horizons (Table 5). Its pattern is controlled by podzolization.

Open-system mass transport functions τ are reported as a function of soil depth in Table 6 and in Figures 4 and 5. Two important assumptions concerning elemental components are required in mass balance calculations using immobile elements (White, 1995; Egli et al., 2001). The first involves the determination of the composition of the parent material. For soils developed at sites on bedrock, the potential errors are confined to local heterogeneities in bedrock composition. The estimation of the initial composition becomes more difficult when soils have developed on sedimentary parent materials such as cover beds. For such deposits, the least weathered horizon in the soil profile is assumed to be the parent material. Consequently, the gains and losses are calculated in relation to the lowermost and least weathered soil horizon. Element losses are

298 observed in all upper horizons of profiles KH1 and KH4. The most intense leaching was found in
299 the E horizon for Fe, Mn, Ca and Mg (Fig. 4 and 5). Also in the Bh horizon, the values were often
300 negative (except for Fe in KH1). Negative mass balance values of up to 75% were found for Mn
301 and Ca in the eluvial horizons of profiles KH2 and KH3. However, an enrichment of P, Al, K and
302 Na (Fig. 4; Table 6) was measured in the uppermost horizons.

303 The highest (K+Ca)/Ti ratios were measured in the middle part of the slope (Table 5). In
304 profiles KH2 and KH3 the ratio decreased with depth, whereas in profiles KH1 and KH4 it was
305 higher in the topsoil and the BC horizon and lowest in the Bh horizon.

306

307 4.4. X-ray diffraction of the clay fraction

308 The investigated clay fraction ($<2\ \mu\text{m}$) is dominated by quartz, mica, vermiculite, kaolinite,
309 chlorite and interstratified minerals (Table 7, Fig. 6). Smectite or interstratified mica-smectite
310 minerals were also detectable. The main difference in soil mineralogy between the soil horizons (E,
311 Bs/BC) is reflected in the secondary phyllosilicate composition (Table 7). As an example, the Mg-
312 saturated clay sample from the E horizon of soil profile KH2 exhibited distinct peaks at 1.43, 1.16,
313 1.01 and 0.71 nm (Fig. 7). Furthermore, small peaks near 2.06 and 1.04 were detectable. Following
314 EG-solvation, the peaks (or part of them) at 1.43 and 2.06 nm shifted to lower angles at 2.30 and
315 1.65 nm. Potassium saturation caused the peak at 1.43 and 1.16 nm to contract to 1.12 and 1.01 nm.
316 After heating at 335 °C, a dominant peak at 1.01 nm was measured. A very minor component
317 persisted at 1.4 nm even after heating at 550 °C. In this sample, two low-charge expandable
318 minerals could be indentified. The peak at 1.65 nm (after EG-solvation) can be attributed to the
319 d_{001}^* of smectite. The shift of the 2.06 nm peak (Mg-saturation) to 2.30 nm (EG-solvation) is a
320 regularly-interstatified mica-smectite (with a d_{002}^* near 1.2 nm). In addition, mica was also present
321 (peak at 1.01 nm). Possibly, a minor proportion of interstratified chlorite-smectite could have
322 contributed to the large shoulder between 1.41 and 1.65 nm (similar to that in the other surface

323 samples). Clear evidence is, however, missing (no change in the 0.71 nm peak after EG solvation;
324 the peak near the 0.498 nm was not investigated; Hubert et al., 2009). The Mg-saturated clay
325 sample from the BC horizon of profile KH2 exhibited an XRD pattern with peaks centered at 1.45,
326 1.27, 1.02 and 0.72 nm. With EG-solvation, a peak at 1.67 nm was detectable. A portion of the 1.45
327 nm peak shifted towards 1.02 and 1.12 nm with heating treatment at 335 °C (Fig. 7). Another part
328 remained near 1.4 nm. Subsequent heating treatment at 550 °C did not noticeably change the XRD
329 pattern. The peak at 1.27 nm (Mg-saturation) could therefore represent an interstratified HIV-mica
330 and/or vermiculite-mica. Chlorite was present in a significant concentration due to the persistence
331 of a peak at 1.4 nm even after the heat treatments. After EG-solvation, smectite was again identified
332 by the peak at 1.67 nm.

333 Pronounced and sharp XRD peaks were denoted at 0.71 or 0.72 and 1.01 nm for the other samples
334 from the E, Bh and BC horizons. The peak at 1.01 nm was attributed to mica (illite), while those at
335 0.71 and 0.72 nm were assigned to kaolinite (the 0.71 nm peak disappeared after heating at 550°C)
336 and chlorite (persistence of the 1.42 nm position after heating at 550°C in some samples; data not
337 shown). In the uppermost soil horizon, chlorite was mostly absent (no or almost no peak around 1.4
338 nm after heating at 550 °C). With increasing soil depth, the peak near 1.4 nm (550 °C) was more
339 pronounced; and subsequently, the proportion of chlorite increased with increasing soil depth (Fig.
340 7). The uppermost horizons all showed some smectite with peaks after EG-solvation near 1.66 to
341 1.70 nm (Fig. 8). Particularly in the E horizon of profile KH1, two swelling phases were present,
342 indicated by a regularly-interstratified mica-smectite (indicated by a peak at 2.67 nm (d_{001}^*) and
343 1.34 nm (d_{002}^*); Fig. 8) and smectite (1.70 nm). The proportion of mica and smectite in the
344 regularly-interstratified mica-smectite was close to a 1:1 ratio (Moore and Reynolds, 1997).

345 For all subsoil samples, a peak at 1.66 to 1.70 nm could be modelled after EG-solvation (Fig. 9).
346 This indicates that some smectite is also present in the subsoil or parent material. Besides mica,
347 kaolinite and chlorite, and also some regularly-interstratified mica-HIV or mica-vermiculite were

348 detectable (Table 7, Fig. 9). In the lower horizons (Bs, BC), a broad peak near 1.4 nm was detected
349 (Fig. 9). Since a small part of this peak shifted to 1.01 nm after K saturation, a minor amount of
350 vermiculite is present. In some cases traces of amphibole (0.85 nm) were also detectable (Fig. 9).
351 The soils at a convex and planar slope position and at the mountain ridge showed a considerable
352 amount of smectite in the topsoil. At these positions, the element leaching rates were elevated
353 compared to profile KH3.
354 The XRD patterns in the d_{060} are presented in Figure 10. X-ray profile fitting enabled the separation
355 of a quartz peak near 0.1544 nm and trioctahedral species at 0.155 to 0.156 nm and 0.1539 to
356 0.1541 nm that can be attributed to biotite and chlorite, respectively (Moore and Reynolds, 1997). A
357 small peak was discernible at 0.1524 nm due to Fe-rich dioctahedral minerals (Fanning et al., 1989);
358 the other dioctahedral species were represented by the peaks in the 0.1515 – 0.1490 nm region. The
359 peak near 0.1491 nm seems to represent kaolinite. The analysis of the d_{060} region shows that the
360 proportion of dioctahedral was higher in the topsoil compared to the subsoil. Subsequently,
361 trioctahedral species decreased when approaching the soil surface.

362

363 4.5. DRIFT measurements

364 The peaks at 375 cm^{-1} , 915 cm^{-1} , 3620 cm^{-1} and 3694 cm^{-1} (Farmer, 1974) confirmed the
365 presence of kaolinite in all soil horizons (Fig. 11). Kaolinite seems to slightly increase towards the
366 soil surface (Fig. 11c). Besides minerals, organic functional groups were detected, such as
367 vibrations of methyl ($-\text{CH}_3$) and methylene ($=\text{CH}_2$) aliphatic compounds ($2900 - 2930\text{ cm}^{-1}$ in the
368 AE and E horizons). Additional organic components were carboxylic groups ($-\text{COOH}$), aromatic
369 compounds and conjugated ketones (1620 cm^{-1}), and the C-O stretching of polysaccharide observed
370 at 1080 cm^{-1} (Stevenson, 1994; Senesi et al., 2003). Weakly developed Si-O stretching bands
371 around 1030 cm^{-1} were attributed to newly-forming dioctahedral phases (Mavris et al., 2011) and
372 the vibration found at circa 1020 cm^{-1} was typical for trioctahedral sheet silicates. Quartz was

373 identified by the doublet peaks at 780 and 800 cm^{-1} (Farmer, 1974). Chlorite was detected around
374 750 cm^{-1} in profiles KH2 and KH4. The peaks around 474 and 531 cm^{-1} corresponded to illite or
375 muscovite (Farmer, 1974; Fig.11).

376

377 **5. Discussion**

378

379 The thickness of the diagnostic horizons and the distribution of Fe and Al in the soil profiles
380 along the catena showed that leaching of solutes occurs vertically and horizontally. Although the
381 mass balance calculations have to be considered with some caution, because they could only be
382 related to the least weathered horizon, they nonetheless show that vertical chemical leaching is most
383 intense on the ridge position. Along the slope (particularly well detectable in the KH3 soil profile,
384 where several elements such as K, Na and Al show an accumulation) an increasing displacement as
385 solutes or in solid form (erosion and accumulation) occurs.

386 The vertical and lateral directions of podzolization influence not only the Al and Fe
387 distribution, but also the fluxes of almost all major and minor elements. The open-system mass
388 transport functions showed a high depletion of Mg and Ca in the E horizons (Fig. 4), giving rise to
389 low concentrations of these elements in the topsoil (Table 5). The mass-fluxes were, to some extent,
390 in relation to the local relief and slope position. The lowest Ca and Mg depletion or even
391 occasionally an accumulation were found in the B horizons of profiles KH2 and KH3: an effect that
392 is due to vertical leaching and lateral mass fluxes. Lateral fluxes are particularly expressed for Al
393 (Fig. 5): a positive mass flux can be measured for KH2 and KH3 throughout the profile. Less
394 negative elemental fluxes or even accumulation seem to be related to undulating or concave surface
395 forms (Table 1, Figs. 4 and 5, Table 7). Negative mass fluxes of Al in upslope soils and positive
396 fluxes along the slope are in agreement with the findings of Sommer et al. (2000; 2001) and the
397 general principle of lateral podzolisation. This model assumes that labile Fe and Al forms (e.g.,

398 complexed to organic ligands) are mobilized in the upper slope section, transported by intra-layer
399 solution flow, and accumulated along and in the lower slope section, which produces thick illuvial
400 horizons in the lower slope.

401 Not only lateral solute fluxes but also erosion of solids, accumulation and mixing due to
402 processes in slope beds (Johnson, 1993) contribute to this configuration. In addition, an aeolian
403 contribution could also seem to have affected the soils (this is discussed below).

404 The vertical distribution of primary minerals, weathering and leaching conditions determines
405 the set of secondary phyllosilicates in the clay fraction and their differentiation along the soil
406 profile. The dominance of illite and the lack of smectite in the clay fraction of Podzols developed
407 from igneous and metamorphic rocks in the Sudeten Mountains has usually been explained by the
408 high abundance of primary mica in the parent materials, the acidic conditions and the very intense
409 chemical weathering (Bogda et al., 1998, Weber, et al., 2012). Kabala (2005) hypothesised that the
410 predominance of illite in the E horizon is owing to two contemporary mechanisms: (1) the physical
411 weathering of coarse feldspar minerals and the subsequent increase in low-dispersive, primary
412 forms of this mineral in the fine silt and clay-size fractions, and (2) the chemical weathering of
413 primary minerals under acidic conditions and the rapid leaching of the weathering products to the B
414 horizons that resulted in a permanent deficiency of alkaline cations in the E horizon giving rise to
415 the hindrance of smectite formation. The presence of (regularly) interstratified mica-smectite,
416 particularly at the shoulder position (KH1), indicates that low-charge expandable phases are
417 actively formed in the soils, because interstratified mica-smectite can be considered a transition
418 mineral between mica and smectite.

419 A dominant phyllosilicate in the clay fraction in the topsoil is, however, mica. The relatively
420 high proportion might be due to passive enrichment caused by some aeolian silt input or by the
421 above-described weathering mechanism (Kabala, 2005). Chlorite is partially dissolved or
422 transformed (this causes, among others, a decrease in the peak at 0.72 nm; Figs. 7 and 8). The

423 influence of aeolian substrates can be inferred from the quantity of silt in the soil and its
424 mineralogical composition. The dominant minerals of aeolian silt are usually quartz (Ollier, 1969;
425 Bronger and Heinkele, 1989), mica and kaolinite. All investigated soils contain 30 – 50% silt in the
426 fine earth fraction. A particularly high silt content is detectable in soil KH3. In all investigated soils,
427 a relative increase in mica (most likely illite) in the clay fraction from the sub- to the topsoil was
428 observed. Particularly at KH3, K and Al are accumulated in the topsoil. This indicates aeolian input.
429 Aeolian material that has been eroded along the slope was probably deposited (locally) at KH3 in
430 the accumulation and mixing process zone (Fig. 12). Furthermore, all soils contain quartz, mica and
431 kaolinite throughout the profile – thus, an aeolian input seems to be likely. However, there are still
432 no convincing data about large-scale aeolian silt enrichment in the subalpine soils of the
433 Karkonosze Mts (Waroszewski et al., 2013).

434 According to Bogda and Chodak (1986), Podzols formed on gneiss or schist bedrock are
435 characterized by the presence of chlorite that may evolve into vermiculite. At least part of the
436 vermiculite is saturated with hydroxy-Al-polymers (Kabala, 2005), giving rise to the formation of
437 HIV. However, the presence of pure smectite has never been confirmed in the clay fraction of
438 Podzols in the Sudetens. Bogda et al. (1998) assumed that the low concentrations of Ca and Mg in
439 the soil solution and the preferential weathering of mica to vermiculite were the causes. Traces of
440 expandable phases may be detectable only in soils formed from fine-grained colluvial² materials in
441 the foot-slope zones, where the addition of an allochthonous aeolian admixture also cannot be
442 excluded (Szerszeń, 1974; Bogda et al., 1998).

443 Our investigations showed that smectites and interstratified mica-smectite minerals can be
444 present even in the eluvial horizon of Podzols, where, according to previous studies, the strong
445 leaching should promote illite formation (Bogda et al., 1998). In a next step, illite – through the

² fine-grained material, often having a silty loam texture, that has been moved down the slope via natural near-surface processes (e.g. erosional wash) and accumulated in concave forms of the landscape

446 release of potassium from interlayers (Bain et al., 1990; Bonifacio et al., 2009) – is transformed into
447 vermiculite. Both mica and chlorites are present in the regolith, thus providing the primary
448 substratum for the formation of smectite and interstratified mica-smectite minerals. Although not
449 clearly identifiable, it seems that some interstratified chlorite-smectite might be present (Hubert et
450 al., 2009), which would indicate an active transformation of chlorite into smectitic minerals. Egli et
451 al. (2003a) demonstrated that strong eluviation, together with the presence of organic ligands,
452 enhanced the removal of Al and enabled smectite formation (even under acidic conditions). At least
453 partially, this seems also to happen in the Podzols of the Kowarski Grzbiet. In addition, amphiboles
454 are highly susceptible to weathering (Proust et al., 2006), especially under intensive weathering and
455 acid conditions (in E horizons). They may also act as the source of smectite formation (Dreher and
456 Niederbuddee, 2000; Egli et al., 2003a). Smectites can be considered to be an end-product of
457 trioctahedral mica (Egli et al., 2008) and chlorite alteration (Righi et al., 1999) with a transition
458 product of irregularly or regularly interstratified 2:1 clay minerals (Egli et al., 2003a; Böhlert et al.,
459 2011). Interstratified mica-smectite is a transitional weathering product that may further evolve into
460 smectite, both dioctahedral and trioctahedral (Churchmann, 1980; Wilson, 1999). Within the
461 investigated soil profiles, a transformation of trioctahedral phases to dioctahedral was discernible
462 from the subsoil to the topsoil. The decrease in trioctahedral phases (chlorite and mica) is related to
463 the increase in dioctahedral phases such as smectite and vermiculite (or even kaolinite; Fig. 10;
464 Mirabella and Egli, 2003). The highest abundance of dioctahedral species seemed to be present in
465 the soil at the ridge position (KH1).

466 Erosion, accumulation and mixing processes lead to inhomogeneous slope deposits.
467 Ultimately, these are not very favourable conditions for smectite formation. Particularly in mid-
468 slope concave positions, Al and Fe do not seem to be fully eluviated or they may even be
469 accumulated, which leads to the preservation of chloritic components (Righi et al., 1993; Carnicelli
470 et al., 1997; Egli et al., 2003b). Nonetheless, some smectite was also found in the subsoil.

471 Mechanical transport of clays with percolating water must be excluded, as clay coatings were found
472 neither on ped faces nor in root channels (Table 2). In addition, the requirements for clay
473 translocation were not present, as the soil conditions were too acidic. Thus, an in situ formation may
474 be considered, even in the subsoil, or more likely an inheritance or accumulation from former
475 upperslope erosion. In some cases, the B horizons contained traces of amphiboles (Table 7). The
476 main secondary products of amphibole weathering are trioctahedral vermiculite, Fe-oxyhydroxides
477 or smectites (Proust, 1985; Velde and Meunier, 2008). We assume that a part of the smectites in the
478 B horizons probably developed from amphibole.

479 The general outline of smectite formation along the investigated slopes and profiles is given in
480 Figure 12. Part of the smectite is inherited from the parent material and part of it is actively formed
481 in the soils. Precursor minerals of smectite formation are chlorite and mica. Such a neo-formation
482 seems to take place particularly at the ridge or lower slope (straight) positions. A concave
483 topography provides less favourable conditions, because Al and Fe are not sufficiently leached
484 away.

485 It is noteworthy that in all soil horizons kaolinite was detected in the clay fraction (Table 7, Fig.
486 11). Kaolinite slightly increased towards the E or AE horizon (cf. Fig. 11c). There may be various
487 sources for the kaolinite: i) it is a direct transformation of mica under intensive chemical weathering
488 conditions (Wilson, 1999, 2004); ii) the origin of the traces of kaolinite could also be traced back to
489 the weathering of plagioclase (Egli et al., 2001b). Together with the pronounced Ca losses in the
490 upper part of the soil profiles (Fig. 4), formation of kaolinite through weathering of plagioclase
491 seems to be likely. Leaching conditions, in particular losses of Ca from surface horizons (Table 5),
492 may foster feldspar weathering and kaolinite formation (Meunier and Velde, 1979). Kaolinite is not
493 a very common component of the clay fraction of podzols. Ross (1980) noted how this is in
494 apparent disagreement with thermodynamical equilibrium calculations, and endorsed the view that
495 this contrast is due to the incomplete evolution and non-equilibrium of soils that, in most cases, are

geologically young and develop in rather cool climates. This would contradict the findings of several others (e.g., Weber et al, 2012) who have postulated a low present-day weathering intensity. Other possible explanations for the presence of kaolinite are: iii) aeolian deposition (Ollier, 1969; Bronger and Heinkele, 1989) and iv) inheritance from the parent material. Given that kaolinite occurs in all soil horizons, even in the parent material, at least a part of it is inherited from the Neogene regoliths (formed under subtropical conditions and advanced chemical weathering) and admixed during the slope–cover formation in the Pleistocene period (Dumanowski, 1961; Migoń and Danišik, 2012).

504

6. Conclusions

506

The slope deposits in the eastern Karkonosze Mountains showed a very complex weathering behaviour. Intensive (vertical) podzolization processes characterized by a distinct leaching of most macroelements were prevalent at the shoulder and lower slope position. Along the slope, lateral translocation of solutes also became increasingly important. In addition, aeolian deposits (illite, kaolinite) also contributed to the soil development. All this had consequences for the formation and transformation of soil minerals. In contrast to previous investigations, not only were illite- and vermiculite-type minerals recognized in the clay fraction of the Karkonosze Mountains, but so were smectites and interstratified smectite-bearing minerals. Where vertical transport processes prevailed, the geochemical conditions for smectite formation were favourable. Strong elemental leaching combined with the presence of precursor minerals such as mica and chlorite were the necessary prerequisites for smectite formation. Even along the slope, smectite was found in the subsoil. This could be due to an inheritance from the parent material or cover bed mixing processes and the accumulation of eroded upslope material.

520

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522

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530

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Table 1. General characteristic of sampling sites

Profile	Elevation (m a.s.l.)	Slope aspect (°)	Slope position	Slope shape (curvature)	Vegetation	Soil classification (IUSS Working Group WRB 2014)
KH1	1268	5	upper slope (shoulder)	convex	<i>Vaccinio- Piceetea</i>	Skeletal Albic Follic Podzol (Loamic)
KH2	1258	15	middle slope (back slope)	straight/concave	<i>Vaccinio- Piceetea</i>	Skeletal Albic Histic Podzol (Loamic)
KH3	1215	10	middle slope (back slope)	concave	<i>Vaccinio- Piceetea</i>	Alumic Skeletal Histic Cambisol (Siltic)
KH4	1142	15	lower slope (footslope)	straight	<i>Vaccinio- Piceetea</i>	Skeletal Albic Follic Podzol (Loamic)

Table 2. Morphological features of the soil profiles

Profile	Soil horizon	Depth (cm)	Colour (moist)	Rock fragments	Structure	Consistency (moist)	Boundary
KH1	O	10-0	10YR 2/2	C			
	E	0-7	10YR 5/1	M	SB, MO	FR	C, W
	Bhs	7-25	7.5YR 4-5/6	A	SB, WE	FR	C
	Bs	25-60	2.5 Y 5/6	A	AS/PL, WE	FR	G
	BC	60+	2.5Y 6/6	D	AS-PL, WE	FR/FI	
KH2	O	12-0	10YR 2/1	C			A/C, W
	E	0-18	10YR 5/2-3	A	SB, MO	FR	C
	Bhs	18-30	10YR 4/6	A	SB, WE	FR	G
	Bs	30-45	10YR 6/8	A	AS/PL, WE	FI	G
	BC	45-60	2.5Y 5/6	D	AS/PL, MO	FI/FR	
KH3	O	10-0	10YR 2/1	F			A/C, W
	AE	0-5	10YR 5/2	M	AB, MO	FR	G
	Bh	5-30	7.5YR 3/2	A	SB, WE	FR	C
	Bs	30-55	7.5YR 4/6	A	AS, WE/MO	FI	G
	BC	55-70	10YR 5/6	D	AS/PL, MO	FI/FR	
KH4	O	12-0	7.5YR 3/1				C, W
	E	0-10	7.5YR 5/2	M	AB, MO	FR	G
	Bh	10-20	7.5YR 3/3	A	AS, MO	FR	G
	Bhs	20-40	10YR 3/4	A	AS/PL, MO	FR/FI	G
	BC	40-60	10YR 4/6	A	AS,MO	FR/FI	

Explanations: abundance of rock fragments: V - very few (0–2%), F - few (2–5%), C - common (5–15%), M - many (15–40%), A - abundant (40–80%), D - dominant (>80%); types of soil structure: AB - angular blocky, AS - angular blocky, SB - subangular blocky, PL – platy; grade of development: WE - weak, MO - moderate,; consistency (moist): FR — friable, FI — firm; horizon boundary (distinctness, topography): A - abrupt, C - clear, G - gradual, W - wavy.

Table 3. Typical physical and chemical properties of the investigated soils

Soil Horizon	Depth (cm)	Bd <div>g cm⁻³</div>	pH <div>KCl</div>	C _{org.} <div>%</div>	Ex. Aci. <div>cmol⁺·(kg)⁻¹</div>	BC	Fe _o	Fe _d	Al _o	Al _d	Fe _o / Fe _d	Al _o ⁺ 1/2Fe _o <div>%</div>
Profile KH1 - Skeletic Albic Folic Podzol (Loamic)												
O	10-0	-	2.60	22.5	3.24	4.07	-	-	-	-	-	-
E	0-7	0.83	3.09	2.18	5.56	1.68	0.23	0.31	0.05	0.06	0.75	0.23
Bhs	7-25	1.20	3.52	3.48	3.36	1.33	3.31	3.74	0.33	0.46	0.89	3.31
Bs	25-60	1.39	4.31	0.60	7.80	0.59	2.47	2.65	0.37	0.51	0.93	2.47
Profile KH2 - Skeletic Albic Histic Podzol (Loamic)												
O	12-0		2.83	15.3	9.52	4.91	-	-	-	-	-	-
E	0-18	1.04	3.47	2.35	5.44	1.02	0.67	0.77	0.10	0.46	0.88	0.44
Bhs	18-30	1.11	3.63	1.83	7.48	1.12	2.09	2.47	0.29	0.34	0.85	1.35
Bs	30-45	1.27	3.86	0.99	5.28	0.73	0.79	1.22	0.19	0.23	0.65	0.60
BC	45-60	1.39	4.08	0.62	4.52	0.68	0.47	0.92	0.20	0.24	0.52	0.44
Profile KH3 - Alumatic Skeletic Histic Cambisol (Siltic)												
O	10-0	-	3.20	30.9	7.64	7.12	-	-	-	-	-	-
AE	0-5	0.67	3.56	9.11	11.1	3.84	1.89	1.95	0.33	0.49	0.97	1.28
Bh	5-30	0.90	3.71	6.14	9.12	1.86	1.74	1.81	0.38	0.54	0.96	1.25
Bs	30-55	0.70	4.10	2.37	4.76	1.06	2.66	2.96	0.81	0.91	0.90	2.15
BC	55-70	1.11	4.32	0.95	2.88	0.87	1.50	1.97	0.51	0.58	0.76	1.27
Profile KH4 - Skeletic Albic Folic Podzol (Loamic)												
O	12-0	-	3.05	34.2	6.92	5.93	-	-	-	-	-	-
E	0-10	0.83	3.11	2.62	5.36	0.68	0.23	0.31	0.06	0.08	0.73	0.18
Bh	10-20	0.90	3.40	4.52	9.96	0.60	2.66	2.72	0.26	0.35	0.97	1.59
Bhs	20-40	0.87	3.70	3.09	7.52	0.59	2.02	2.36	0.34	0.49	0.86	1.35
BC	40-60	1.01	4.05	2.12	5.24	0.42	2.07	2.68	0.51	0.60	0.77	1.55

Explanations: Bd — bulk density, C_{org.} — organic carbon, BC — sum of exchangeable base cations, Ex. Aci. — exchangeable acidity, Fe_d, Al_d — dithionite iron and aluminum, Fe_o, Al_o — oxalate iron and aluminum, Fe_o/Fe_d — iron activity index, Al_o+1/2Fe_o — illuviation (podzolization) index.

Table 4. Particle size distribution of the investigated soils

Soil horizon	Depth [cm]	Skeleton content	Sand			Silt			Clay fraction	Sum of fractions		Texture class	
			vcS	cS	mS	fS	vfS	cSi		fSi	Sand		Silt
%													
Profile KH1 - Skeletic Albic Folic Podzol (Loamic)													
E	0-7	10	3	9	10	15	12	20	27	4	49	47	SL
Bhs	7-25	50	3	7	9	19	14	25	19	4	52	44	SL
Bs	25-60	80	6	7	8	14	12	24	23	6	47	47	SL
Profile KH2 - Skeletic Albic Histic Podzol (Loamic)													
E	0-18	50	4	7	9	17	11	23	23	6	48	46	SL
Bhs	18-30	70	5	5	7	13	16	26	24	4	46	50	SiL
Bs	30-45	70	4	5	6	10	17	25	23	10	42	48	SL
BC	45-60	80	4	5	6	12	13	33	17	10	40	50	SiL
Profile KH3 - Alomic Skeletic Histic Cambisol (Siltic)													
AE	0-5	30	2	3	4	9	10	22	46	4	28	68	SiL
Bh	5-30	80	5	8	10	14	11	22	28	2	48	50	SiL
Bs	30-55	80	5	5	7	21	13	30	16	3	51	46	SL
BC	55-70	90	4	7	10	18	19	23	15	4	58	38	SL
Profile KH4 - Skeletic Albic Folic Podzol (Loamic)													
E	0-10	40	5	9	9	19	16	18	20	4	58	38	SL
Bh	10-20	50	3	6	14	22	19	16	16	4	64	32	SL
Bhs	20-40	50	7	7	9	21	19	16	17	4	63	33	SL
BC	40-60	50	8	6	9	25	20	14	12	6	68	26	SL

Textural classes: SL—sandy loam, SiL —silt loam; fine-earth fractions (<2 mm):vcS—very coarse sand, cS—coarse sand, mS —medium sand, fS— fine sand, vfS — very fine sand, cSi — coarse silt, fSi — fine silt.

Table 5. Major element contents in the fine earth fraction

Horizon	Depth cm	Na ₂ O	MgO	K ₂ O	CaO	Al ₂ O ₃	Fe ₂ O ₃	SiO ₂	P ₂ O ₅	MnO	TiO ₂	(K+Ca)/Ti
%												
Profile KH1 - Skeletic Albic Folic Podzol (Loamic)												
E	0-7	1,28	0,85	4,51	0,07	17,75	2,13	67,33	0,026	0,029	1,140	6,79
Bhs	7-25	0,99	1,20	2,88	0,13	13,39	10,75	50,68	0,166	0,058	0,855	5,92
Bs	25-60	1,16	1,65	3,68	0,22	16,19	6,13	65,42	0,050	0,092	0,989	6,62
Profile KH2 - Skeletic Albic Histic Podzol (Loamic)												
E	0-18	0,99	0,88	3,63	0,07	14,52	2,76	67,24	0,105	0,020	1,048	5,97
Bhs	18-30	0,88	1,43	2,93	0,18	12,89	7,48	61,76	0,100	0,044	0,895	5,84
Bs	30-45	0,99	1,47	3,01	0,25	12,72	5,39	68,35	0,042	0,056	0,965	5,65
BC	45-60	1,10	1,47	3,08	0,27	12,70	4,97	71,18	0,016	0,058	1,011	5,54
Profile KH3 - Alomic Skeletic Histic Cambisol (Siltic)												
AE	0-5	1,18	1,12	4,23	0,11	18,34	4,91	45,93	0,422	0,010	0,759	9,64
Bh	5-30	1,05	1,17	3,84	0,06	17,67	5,17	52,65	0,406	0,015	0,865	7,64
Bs	30-55	0,95	1,35	2,68	0,14	14,55	8,24	52,11	0,196	0,029	0,781	6,08
BC	55-70	1,13	1,71	3,20	0,15	16,00	6,63	58,79	0,134	0,050	0,944	5,98
Profile KH4 - Skeletic Albic Folic Podzol (Loamic)												
E	0-10	1,16	0,77	3,14	0,11	13,29	1,97	72,74	0,054	0,020	1,094	5,01
Bh	10-20	1,04	0,93	2,53	0,15	12,04	6,32	57,51	0,148	0,022	0,970	4,64
Bhs	20-40	1,04	0,98	2,45	0,19	11,55	6,26	61,76	0,109	0,025	0,985	4,50
Bs	40-60	1,05	1,47	2,58	0,25	13,82	7,43	58,19	0,125	0,038	0,970	4,87

Table 6. Relative elemental losses for Si, P and Mn (open-system mass transport function $\tau_{j,w}$)

Horizon	Depth (cm)	Si	P	Mn
Profile KH1 - Skeletic Albic Folic Podzol (Loamic)				
E	0-7	-0,10	-0,55	-0,73
Bhs	7-25	-0,10	2,86	-0,27
Bs	25-60	0	0	0
Profile KH2 - Skeletic Albic Histic Podzol (Loamic)				
E	0-18	-0,08	5,22	-0,66
Bhs	18-30	-0,01	5,93	-0,14
Bs	30-45	0,01	1,70	0,02
BC	45-60	0	0	0
Profile KH3 - Alumic Skeletic Histic Cambisol (Siltic)				
AE	0-5	-0,03	2,92	-0,74
Bh	5-30	-0,02	2,31	-0,66
Bs	30-55	0,07	0,77	-0,30
BC	55-70	0	0	0
Profile KH4 - Skeletic Albic Folic Podzol (Loamic)				
E	0-10	0,11	-0,62	-0,53
Bh	10-20	-0,01	0,18	-0,41
Bhs	20-40	0,05	-0,14	-0,33
Bs	40-60	0	0	0

Table 7. Mineral composition of the clay fraction (< 2 µm)

Soil profile	Soil horizon	Quartz	Mica	Chlorite	Kaolinite	Smectite	Interstratified mica-smectite	Interstratified mica-vermiculite (or HIV)	Vermiculite	Amphibole
KH1	E	++	++	-	+	(+)	+	(+)	+	-
	Bhs	++	++	(+)	+	(+)	-	+	(+)	-
	Bs	++	++	-	+	(+)	-	+	(+)	(+)
KH2	E	++	++	(+)	+	+	(+)	(+)	+	-
	Bhs	++	++	+	+	+	-	+	(+)	-
	Bs	++	++	+	+	+	-	+	(+)	-
	BC	++	++	+	+	+	-	+	(+)	(+)
KH3	AE	++	++	(+)	+	(+)	-	+	+	-
	Bh	++	++	(+)	+	+	-	+	(+)	-
	Bs	++	++	+	+	(+)	-	+	(+)	-
	BC	++	++	+	+	(+)	-	+	(+)	(+)
KH4	E	++	++	-	+	+	(+)	+	+	-
	Bh	++	++	+	+	(+)	-	+	(+)	-
	Bhs	++	++	+	+	(+)	-	+	(+)	-
	Bs	++	++	+	+	(+)	-	+	(+)	(+)

++ major, + minor, (+) trace, - absent

Figure captions

Fig. 1. Southern slopes of Skalny Stół (Kowarski Grzbiet/ Střeha) with the investigated soil profiles (KH 1-4) along a catena (view from Czarna Kopa).

Fig. 2. Soils along the studied toposequence.

Fig. 3. Rock thin sections: quartzite (a - PPL) and mica schist (b- XPL, c -PPL ,d -XPL). The individual minerals are: Qz - quartz, Ms - muscovite, Chl - chlorite, Grt - garnet, Tur – tourmaline.

Fig. 4. Open-system mass transport function (τ) for Na, Ca, K, Mg along the soil profiles.

Fig. 5. Open-system mass transport function (τ) for Fe and Al along the soil profiles.

Fig. 6. XRD pattern of the randomly-oriented powder mounts (profile KH1).

Fig. 7. XRD patterns of soil clays ($< 2 \mu\text{m}$) of the Eg and BC horizon of profile KH2. The XRD curves are corrected for Lorentz and polarization factors; d -spacings are given in nm. EG = ethylene glycol solvation, Mg = Mg-saturation, K = K-saturation and corresponding heating treatments.

Fig. 8. XRD patterns of the EG-solvated clay fraction from the topsoils along the toposequence. The measured values (squares), modelled elementary curves and the modelled overall curve are shown. The XRD curves are corrected for Lorentz and polarization factors; d -spacings are given in nm.

Fig. 9. XRD patterns of the EG-solvated clay fraction from the subsoils (B and BC horizon) along the toposequence. The measured values (squares), modelled elementary curves and the modelled overall curve are shown. The XRD curves are corrected for Lorentz and polarization factors; d -spacings are given in nm.

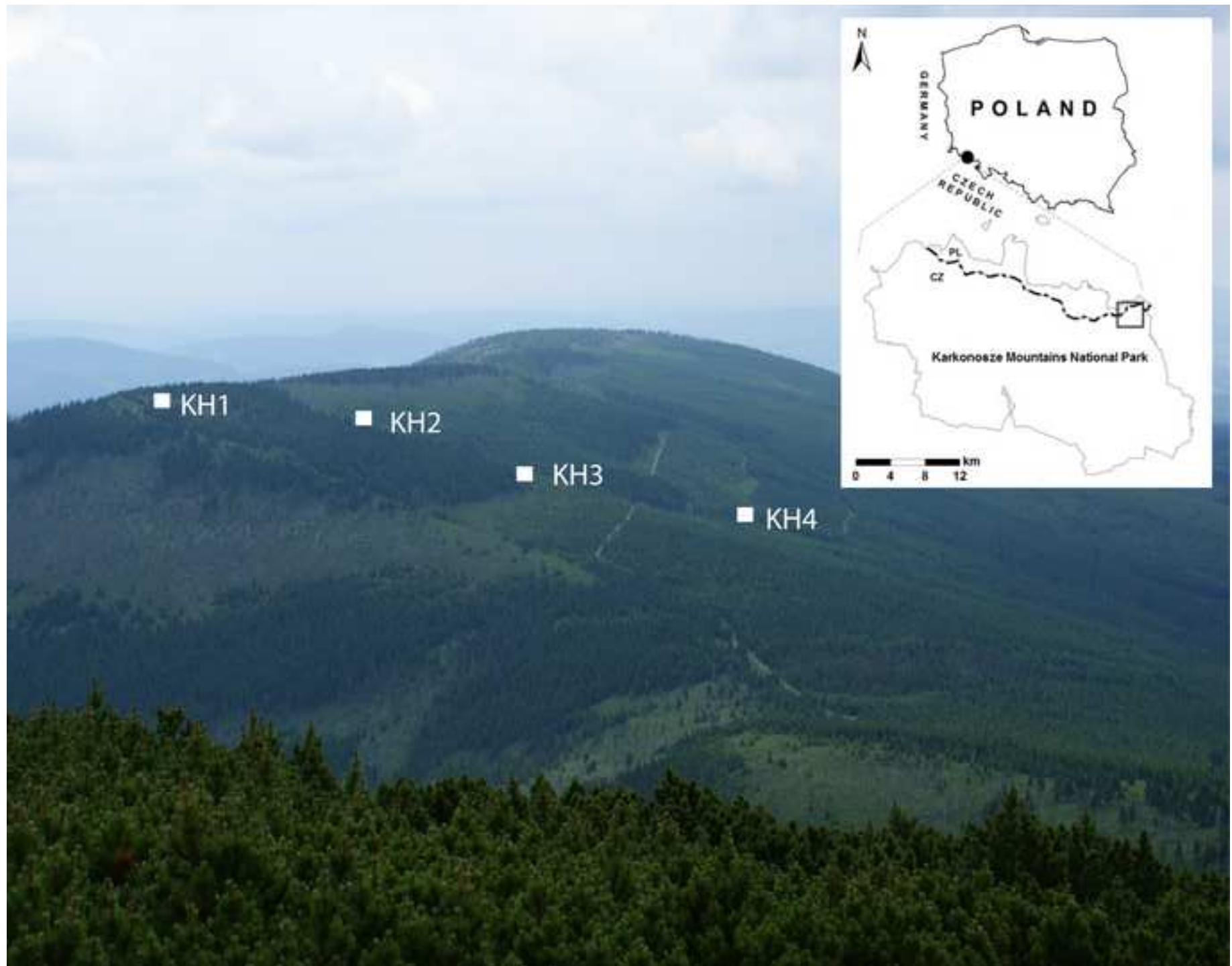
Fig. 10. XRD patterns in the d_{060} region of the soil clays from the investigated sites (topsoil and subsoil). The peak range between 0.1560 and 0.1530 nm was assigned to trioctahedral and the one between 0.1530 and 0.1490 to dioctahedral phases. Given are the measured values (squares), modelled elementary curves and the modelled overall curve.

Fig.11. DRIFT spectra of selected soil samples (clay fraction): a – horizons of profile KH3; b – AE and E horizons of all soils along the catena; c – Normalised (min-max method) IR spectra of soil clays in the OH-stretching region 4000–3000 cm^{-1} .

Fig. 12. Scheme of presumed weathering and mineral transformation processes in the soil profiles and along the slope. Measurable processes are given by solid lines and presumed processes by dashed lines.

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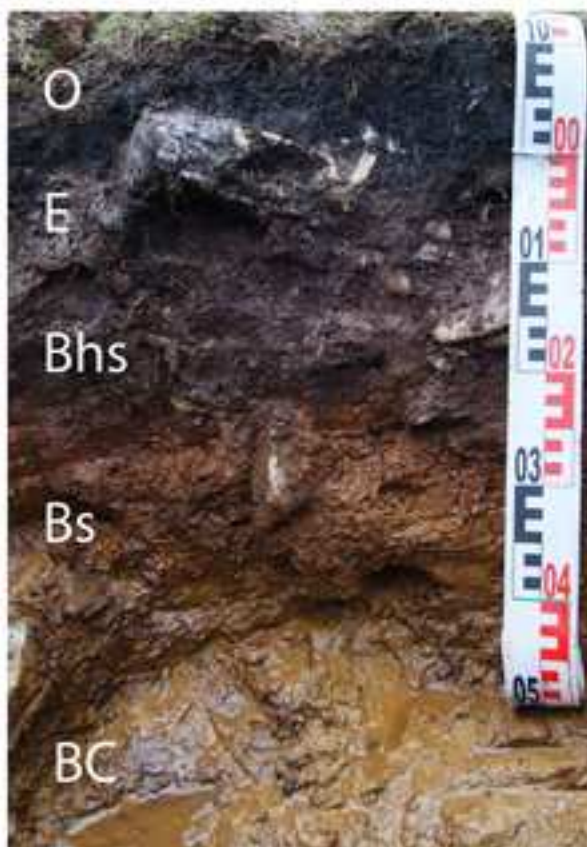
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KH1



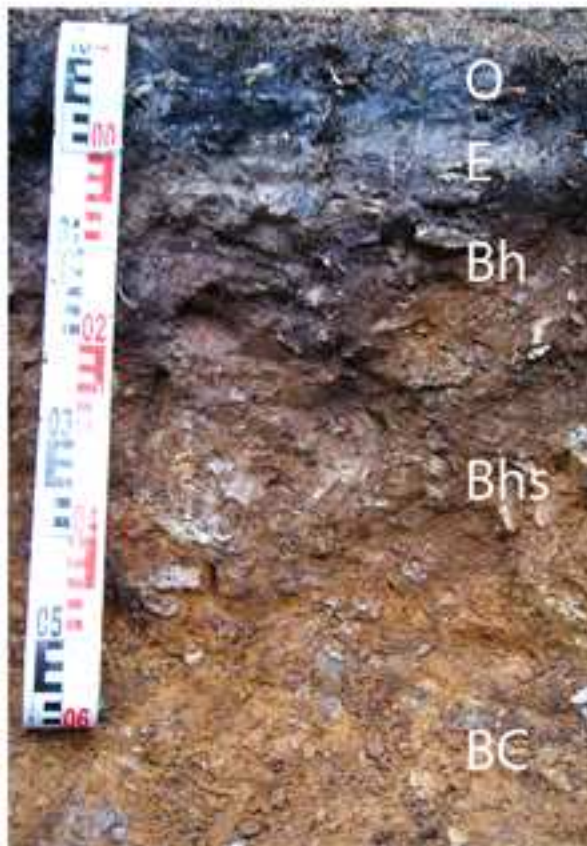
KH2



KH3



KH4



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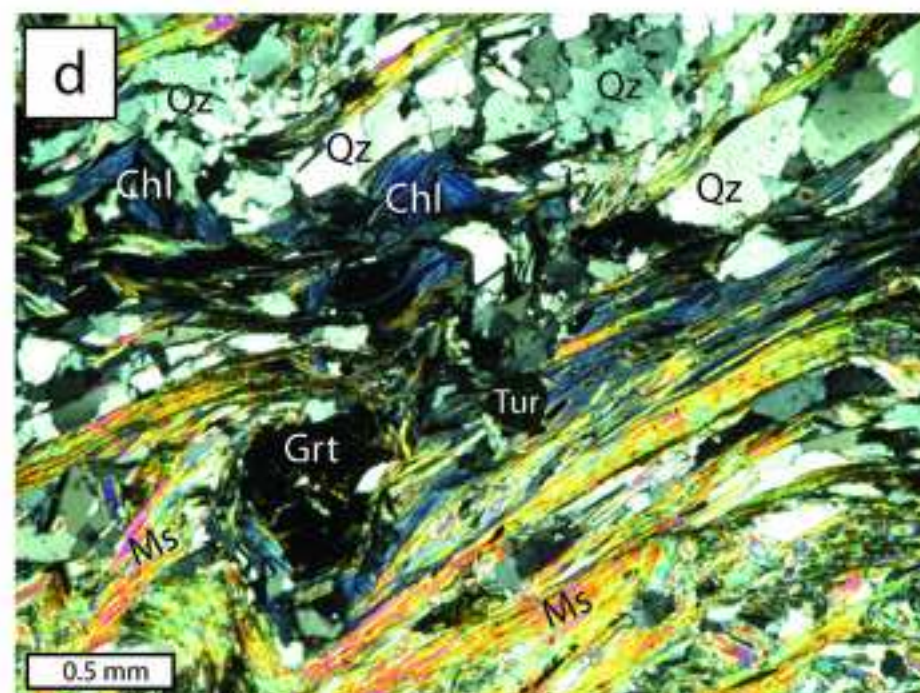
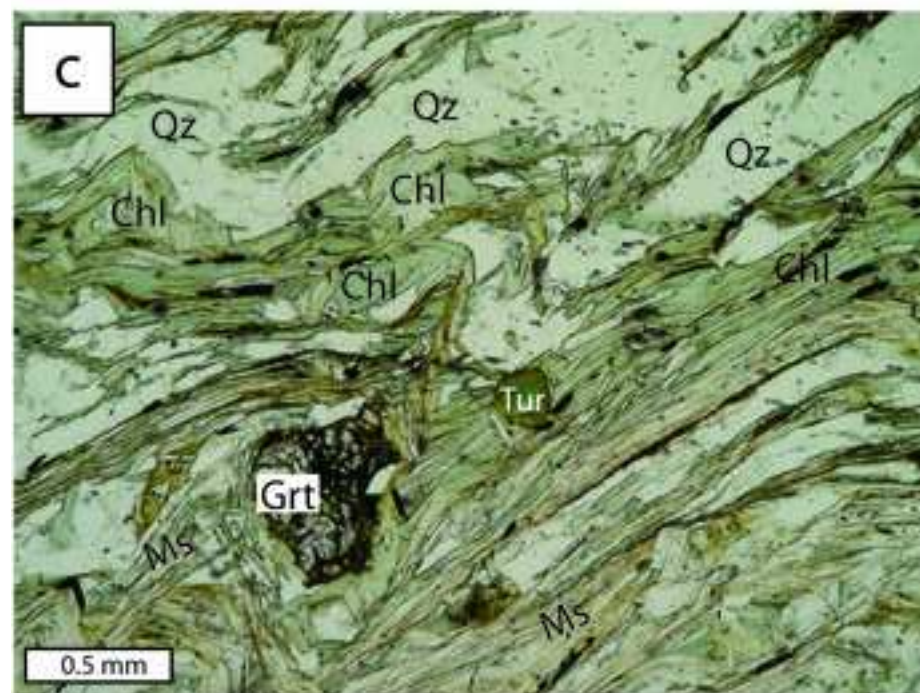
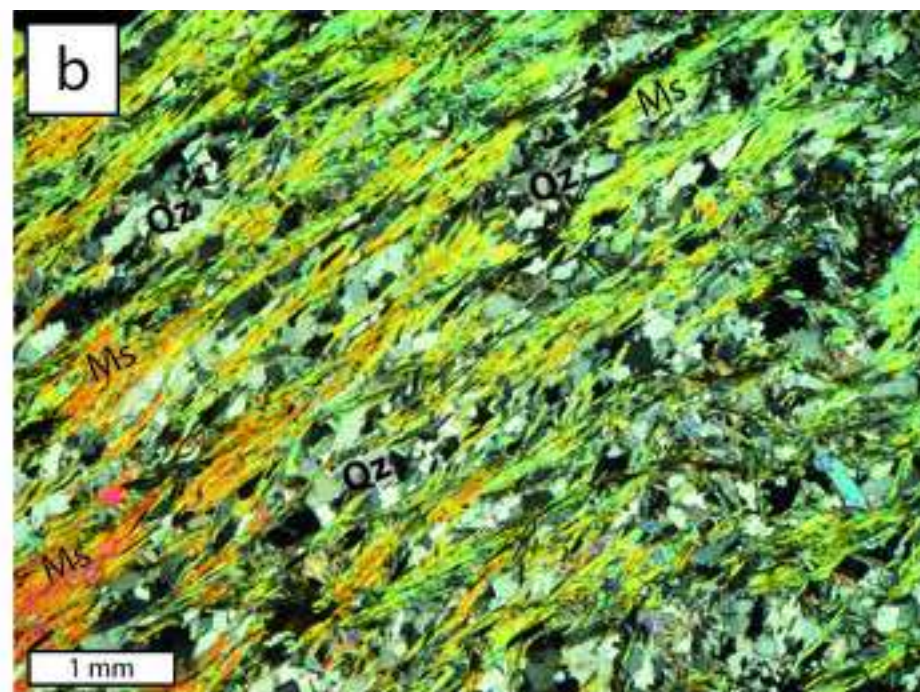
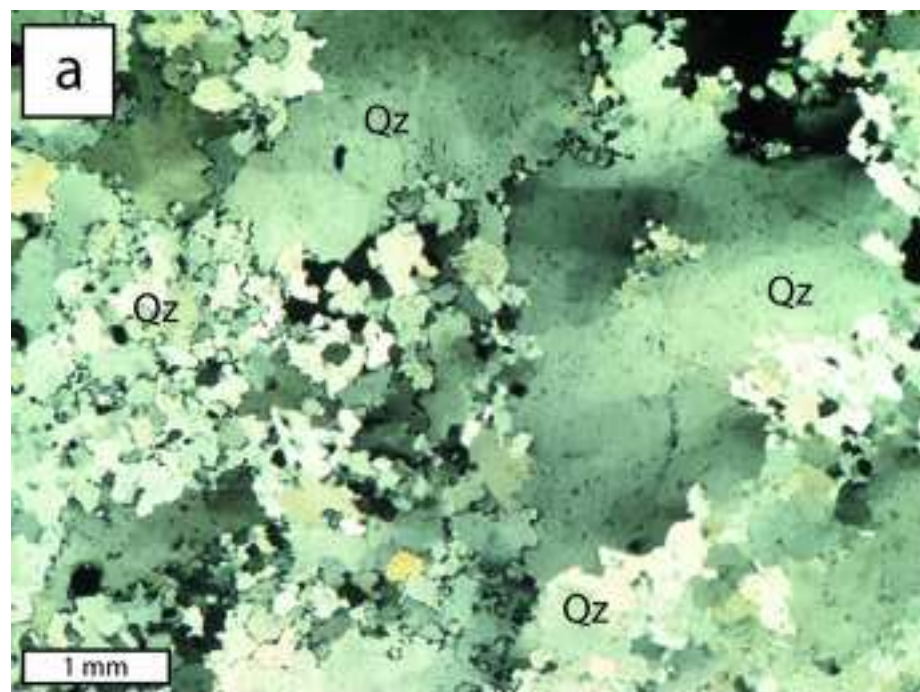


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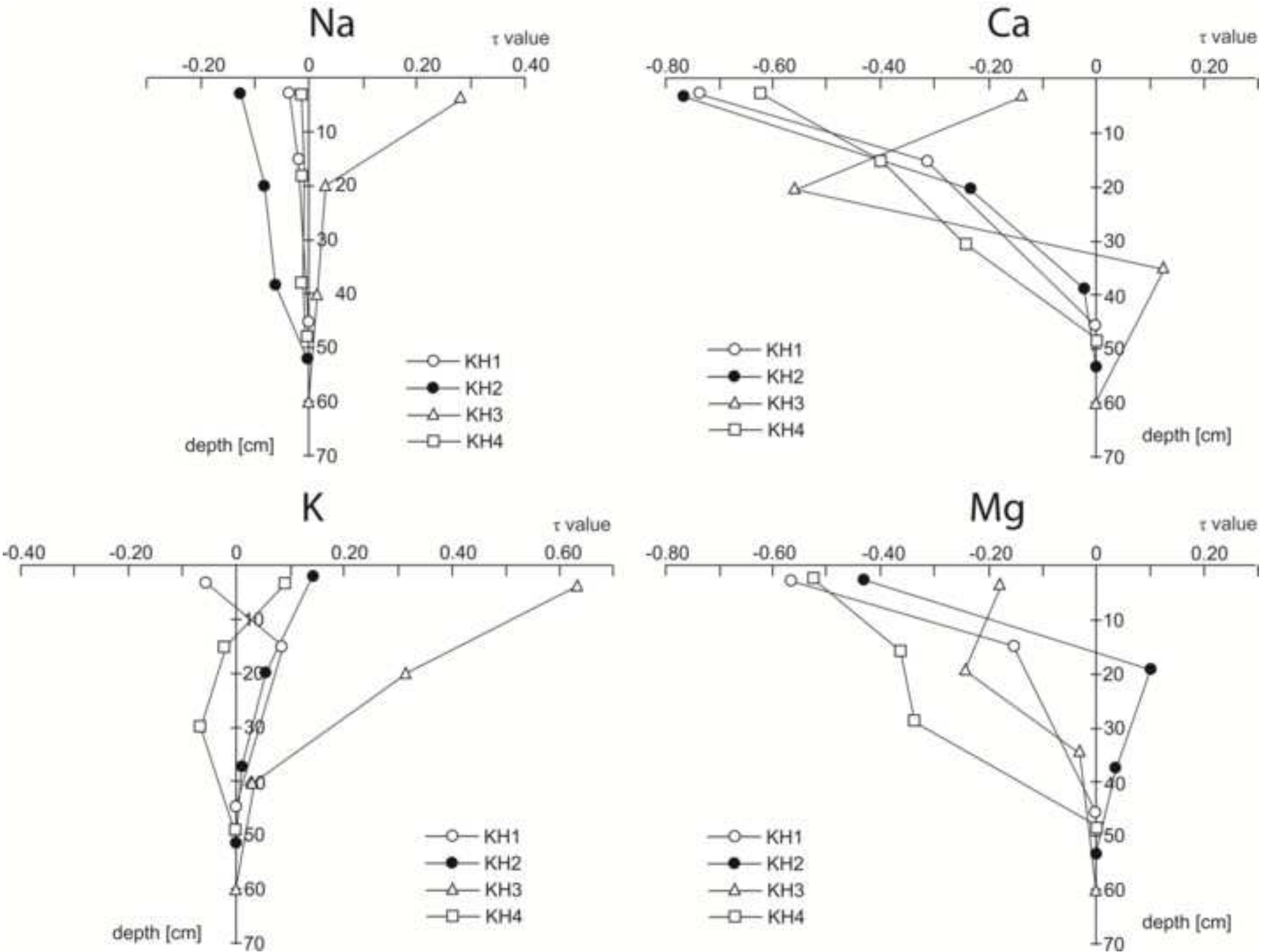


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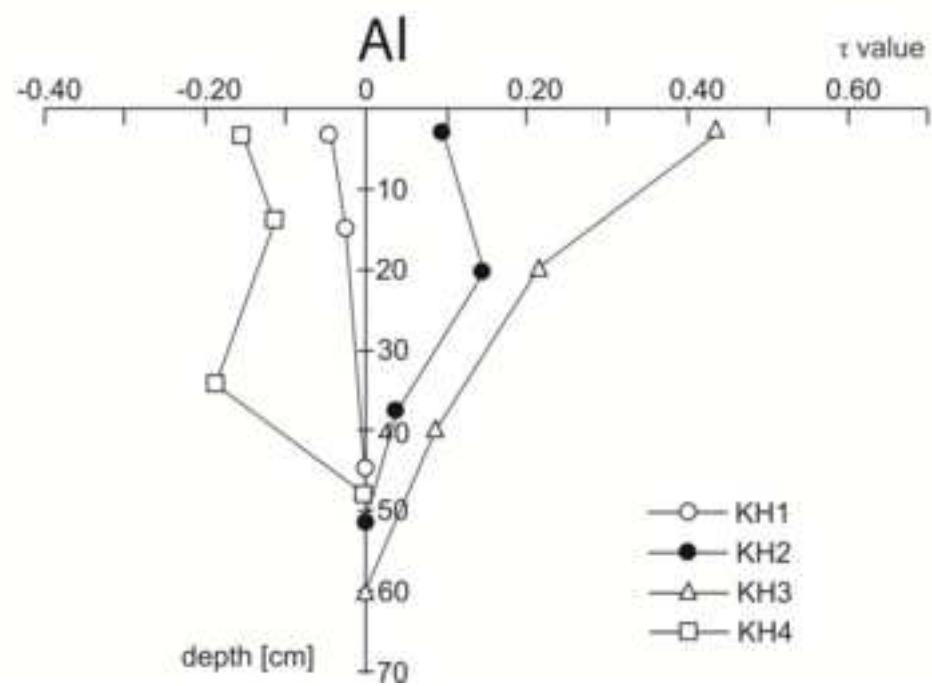
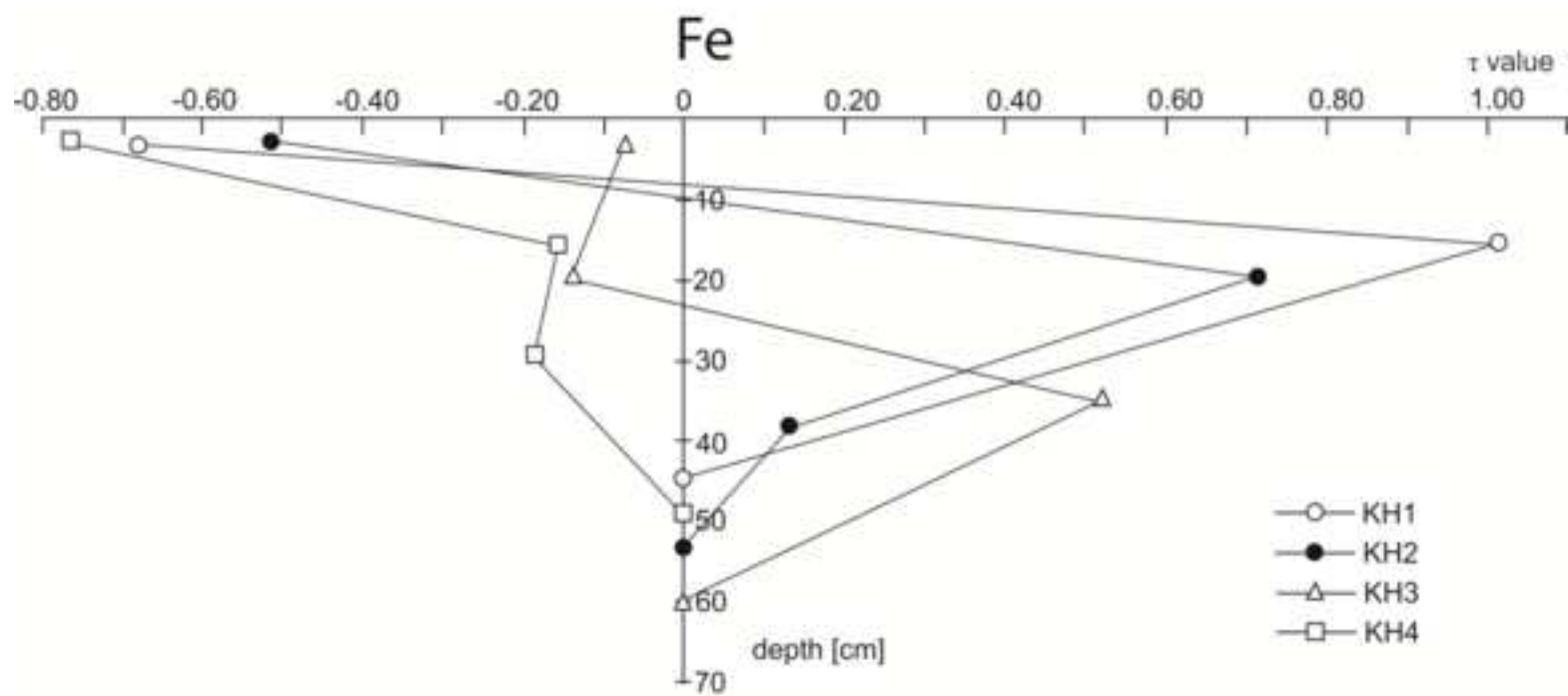
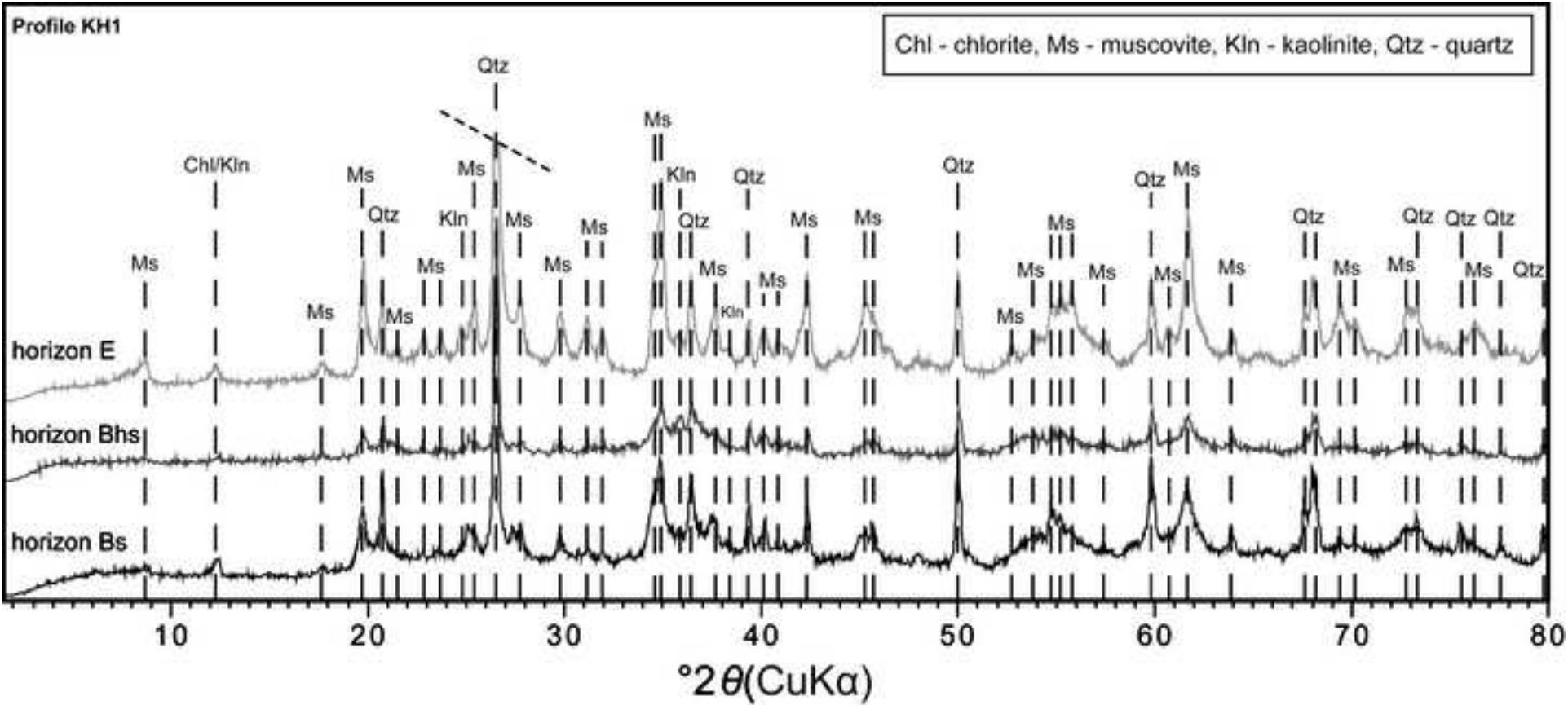


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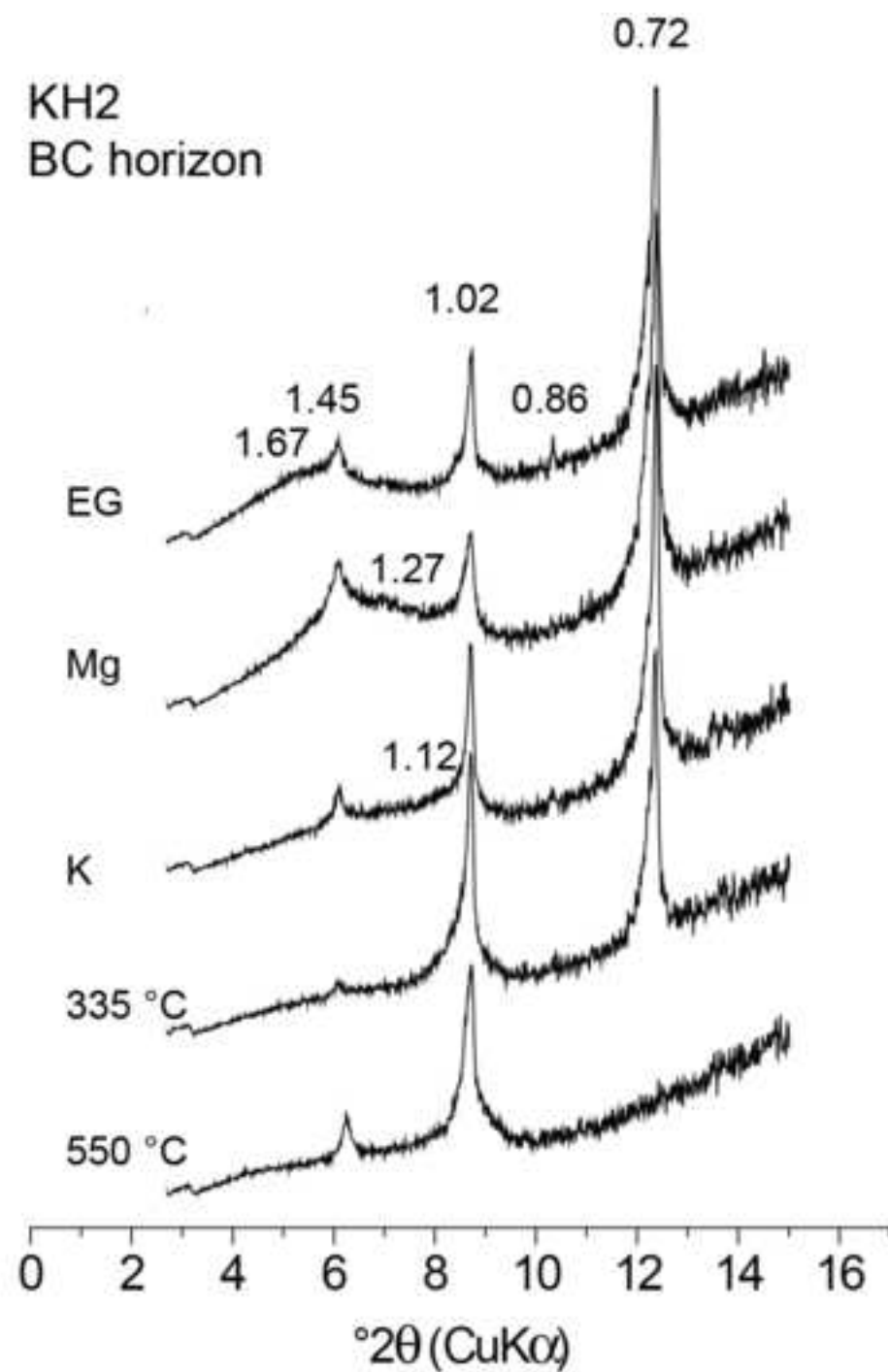
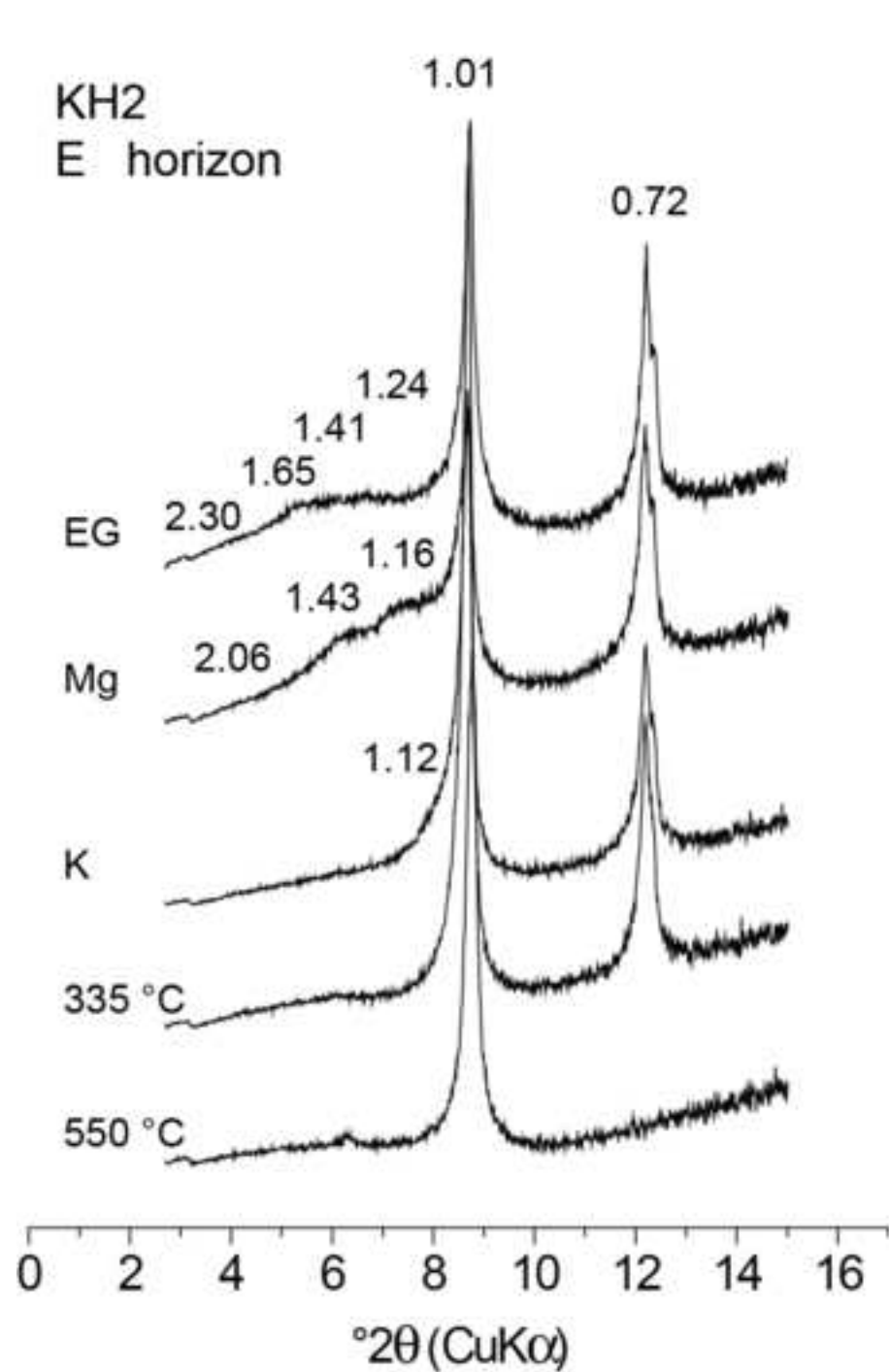
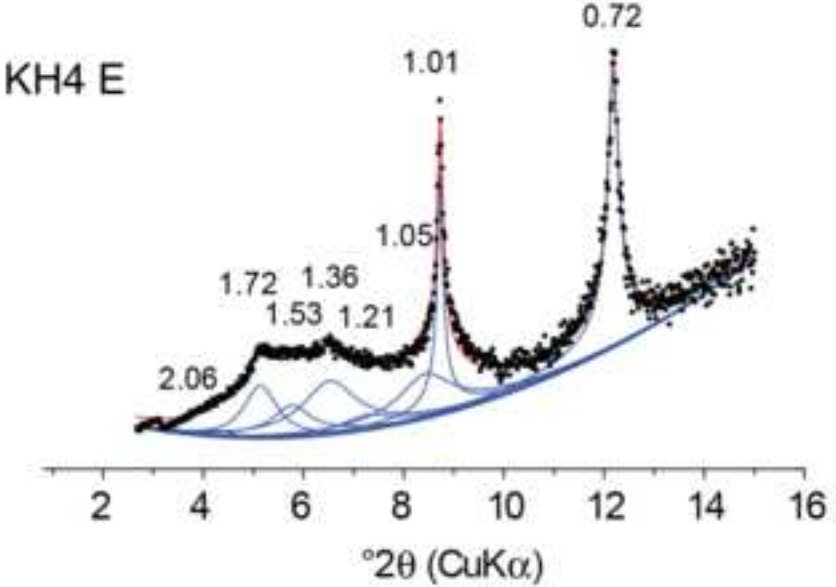
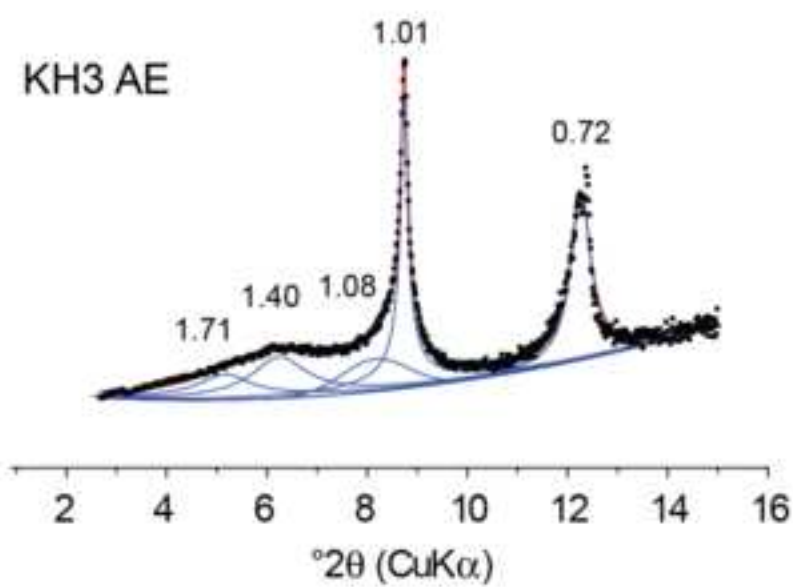
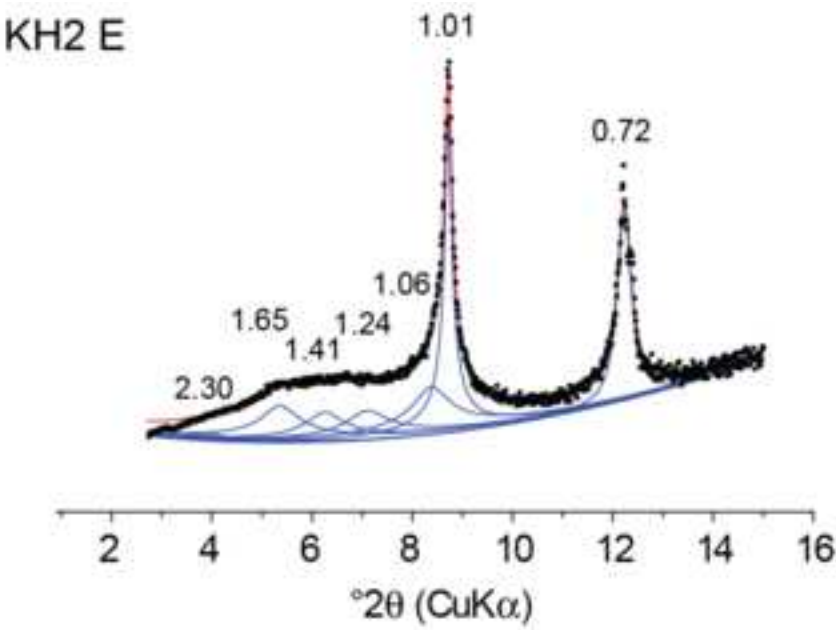
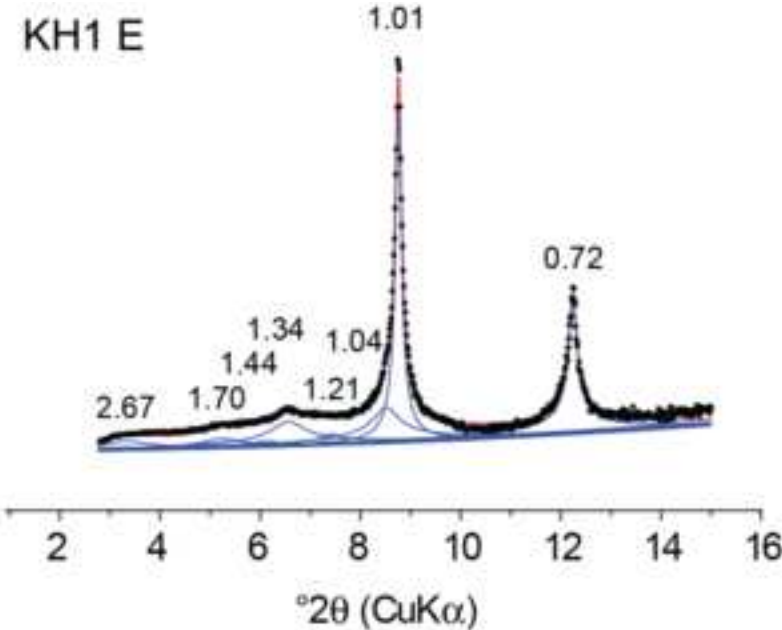


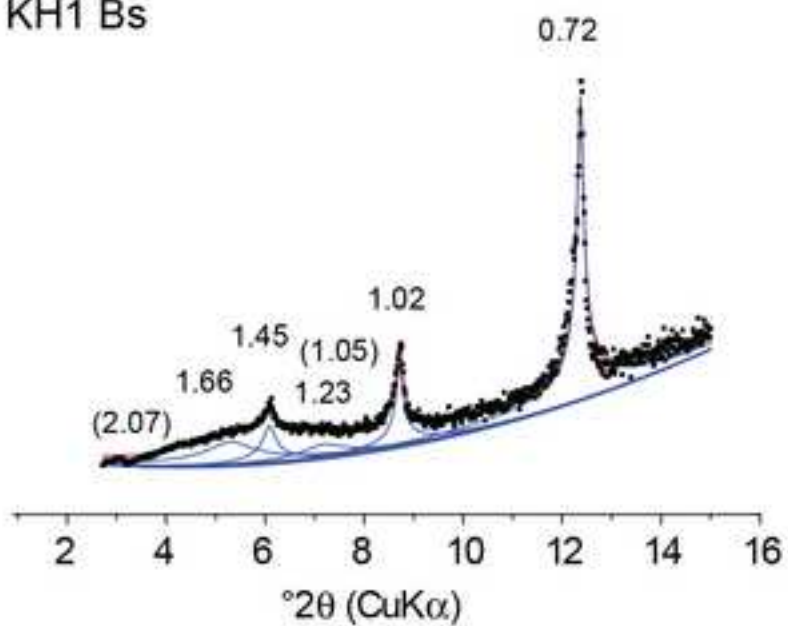
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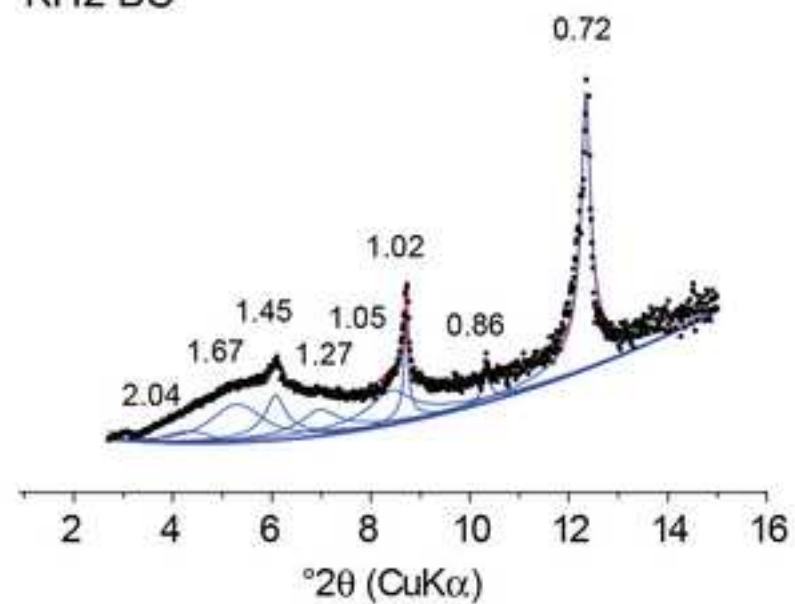
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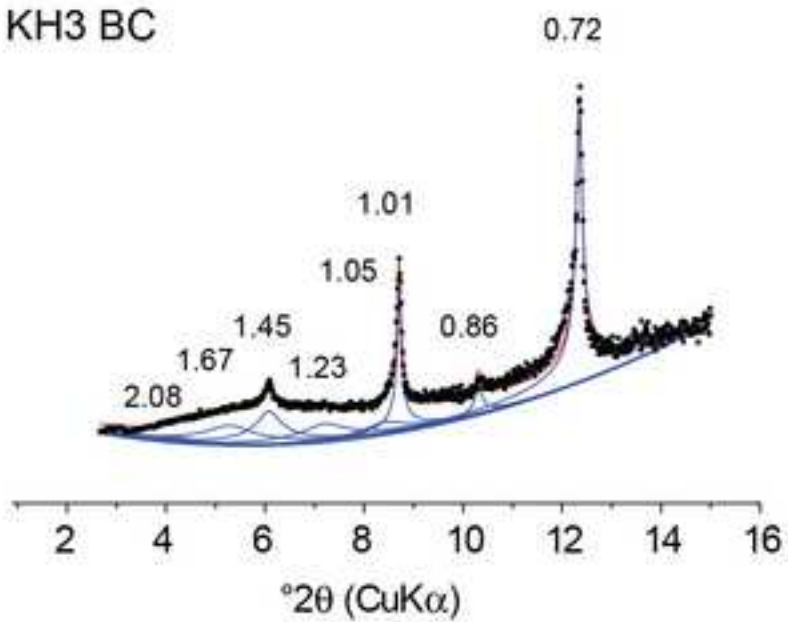
KH1 Bs



KH2 BC



KH3 BC



KH4 Bhs

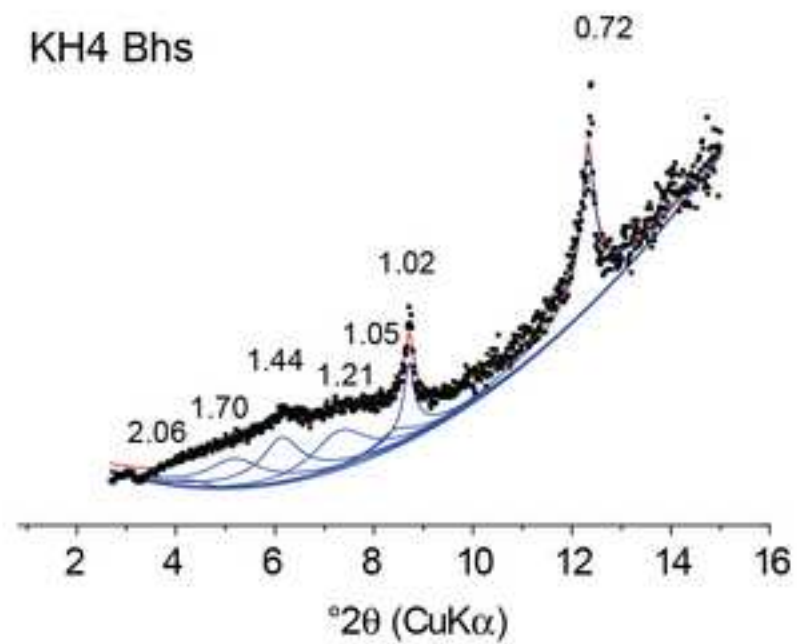


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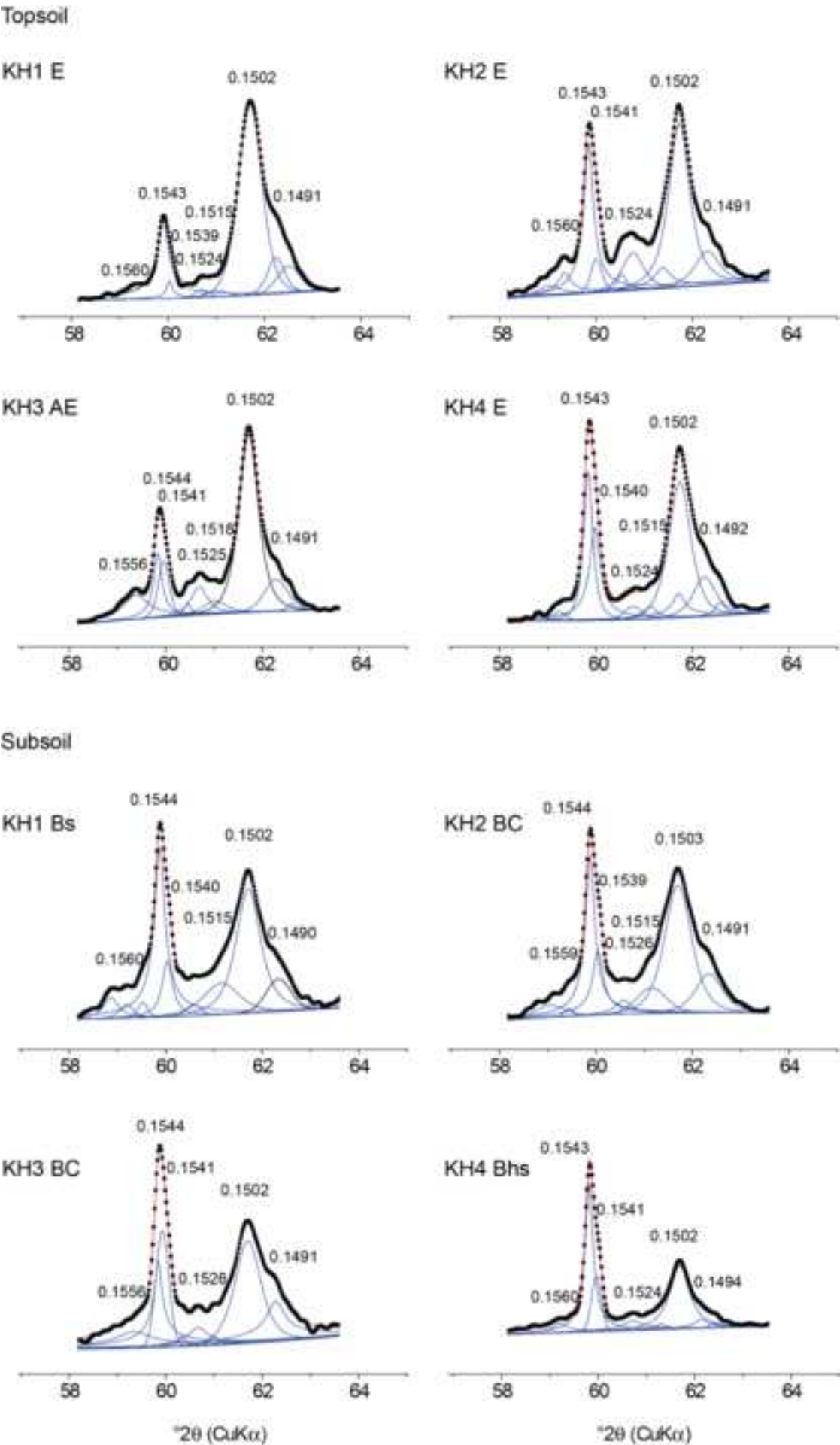


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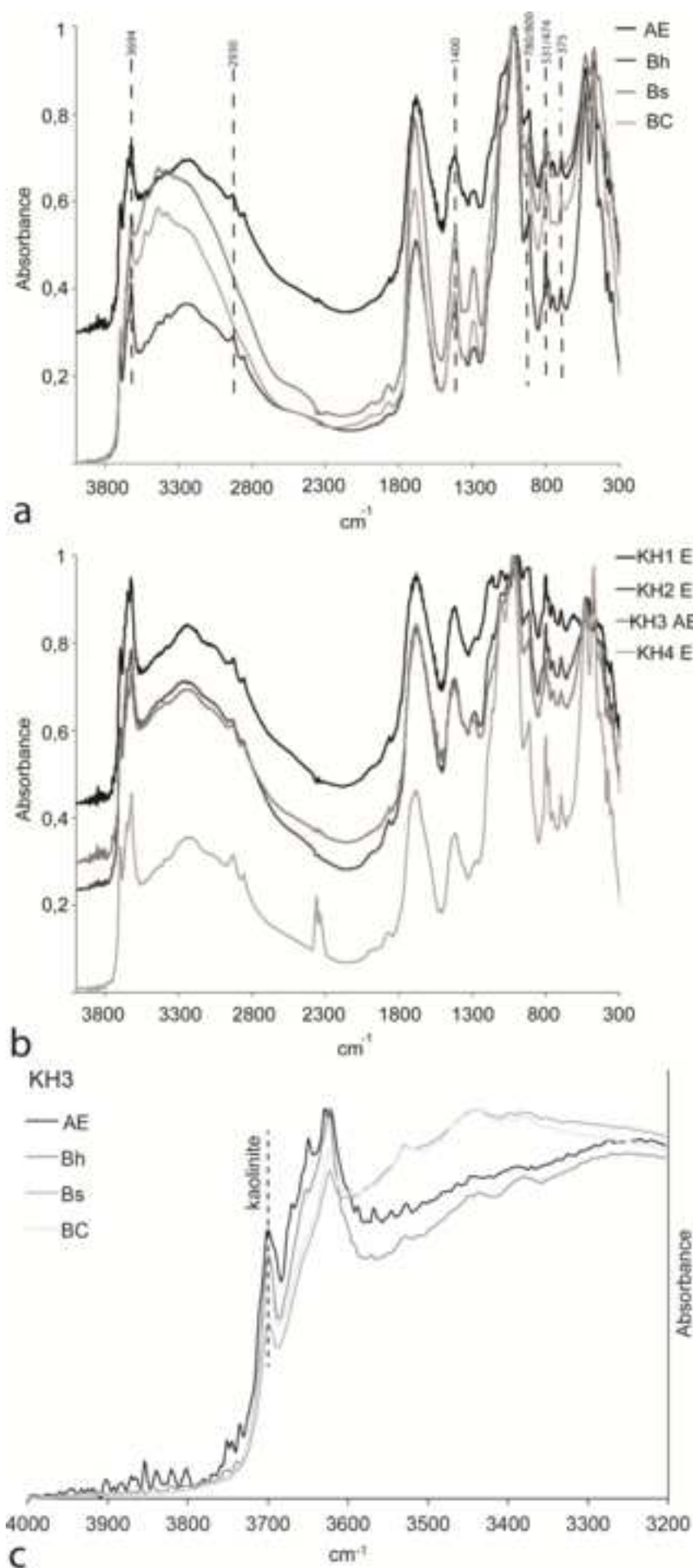


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